

# Neostigmine

<b>Other names:</b>	Ammonium, (m-hydroxyphenyl)trimethyl-, dimethylcarbamate Benzenaminium, 3-(((dimethylamino)carbonyl)oxy)-N,N,N-trimethyl- Eustigmin Eustigmine (m-Hydroxyphenyl)trimethylammonium dimethylcarbamate Prostigmin Prostigmine Vagostigmine 3-[[[(Dimethylamino)carbonyl]oxy]-N,N,N-trimethylbenzenaminium Proserine Synstigmin Synstigmine (m-Hydroxyphenyl)dimethylamine dimethylcarbamate
<b>Inchi:</b>	InChI=1S/C12H19N2O2/c1-13(2)12(15)16-11-8-6-7-10(9-11)14(3,4)5/h6-9H,1-5H3/q+1
<b>InchiKey:</b>	ALWKGYPQUAPLQC-UHFFFAOYSA-N
<b>Formula:</b>	C12H19N2O2
<b>SMILES:</b>	CN(C)C(=O)Oc1cccc([N+](C)(C)C)c1
<b>Mol. weight [g/mol]:</b>	223.29
<b>CAS:</b>	59-99-4

## Physical Properties

Property code	Value	Unit	Source
gf	88.20	kJ/mol	Joback Method
hf	-231.50	kJ/mol	Joback Method
hfus	27.64	kJ/mol	Joback Method
hvap	58.63	kJ/mol	Joback Method
log10ws	3.06		Crippen Method
logp	1.944		Crippen Method
mcvol	185.730	ml/mol	McGowan Method
pc	2365.67	kPa	Joback Method
tb	607.49	K	Joback Method
tc	811.98	K	Joback Method
tf	384.67	K	Joback Method
vc	0.668	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	477.73	J/mol×K	607.49	Joback Method
cpg	494.72	J/mol×K	641.57	Joback Method
cpg	510.74	J/mol×K	675.65	Joback Method
cpg	525.80	J/mol×K	709.73	Joback Method
cpg	539.92	J/mol×K	743.81	Joback Method
cpg	553.10	J/mol×K	777.90	Joback Method
cpg	565.35	J/mol×K	811.98	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C59994&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C59994&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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