

# 4-Undecanone

<b>Other names:</b>	Heptyl propyl ketone undecan-4-one
<b>Inchi:</b>	InChI=1S/C11H22O/c1-3-5-6-7-8-10-11(12)9-4-2/h3-10H2,1-2H3
<b>InchiKey:</b>	NBSLHMOSERBUOV-UHFFFAOYSA-N
<b>Formula:</b>	C11H22O
<b>SMILES:</b>	CCCCCCCC(=O)CCC
<b>Mol. weight [g/mol]:</b>	170.29
<b>CAS:</b>	14476-37-0

## Physical Properties

Property code	Value	Unit	Source
gf	-87.18	kJ/mol	Joback Method
hf	-382.95	kJ/mol	Joback Method
hfus	25.84	kJ/mol	Joback Method
hvap	46.83	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	3.716		Crippen Method
mcvol	167.420	ml/mol	McGowan Method
pc	2045.61	kPa	Joback Method
rhoc	245.22 ± 6.81	kg/m3	NIST Webbook
rinpol	1208.00		NIST Webbook
rinpol	1208.00		NIST Webbook
rinpol	1208.00		NIST Webbook
ripol	1540.00		NIST Webbook
ripol	1475.00		NIST Webbook
ripol	1540.00		NIST Webbook
ripol	1476.00		NIST Webbook
ripol	1475.00		NIST Webbook
ripol	1518.00		NIST Webbook
tb	504.95	K	Joback Method
tc	680.90 ± 1.60	K	NIST Webbook
tf	278.00 ± 4.00	K	NIST Webbook
vc	0.657	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	388.23	J/molxK	504.95	Joback Method
cpg	403.46	J/molxK	533.52	Joback Method
cpg	418.07	J/molxK	562.10	Joback Method
cpg	432.09	J/molxK	590.67	Joback Method
cpg	445.51	J/molxK	619.25	Joback Method
cpg	458.36	J/molxK	647.82	Joback Method
cpg	470.65	J/molxK	676.40	Joback Method
dvisc	0.0044945	Paxs	263.66	Joback Method
dvisc	0.0020210	Paxs	303.88	Joback Method
dvisc	0.0010954	Paxs	344.09	Joback Method
dvisc	0.0006749	Paxs	384.30	Joback Method
dvisc	0.0004558	Paxs	424.52	Joback Method
dvisc	0.0003295	Paxs	464.74	Joback Method
dvisc	0.0002508	Paxs	504.95	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	379.20	K	1.70	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.60833e+01
Coeff. B	-4.76763e+03
Coeff. C	-8.22700e+01
Temperature range (K), min.	384.10
Temperature range (K), max.	524.87

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14476370&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14476370&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rhoc:</b>	Critical density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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