

# Oxetane, 3,3-bis-(chloromethyl)

<b>Other names:</b>	3,3-bis(chloromethyl)oxetane
<b>Inchi:</b>	InChI=1S/C5H8Cl2O/c6-1-5(2-7)3-8-4-5/h1-4H2
<b>InchiKey:</b>	CXURGFRDGRIOIKG-UHFFFAOYSA-N
<b>Formula:</b>	C5H8Cl2O
<b>SMILES:</b>	C1CC1(CCl)COC1
<b>Mol. weight [g/mol]:</b>	155.02
<b>CAS:</b>	78-71-7

## Physical Properties

Property code	Value	Unit	Source
chl	-2898.30 ± 1.80	kJ/mol	NIST Webbook
gf	-75.60	kJ/mol	Joback Method
hf	-204.00 ± 1.90	kJ/mol	NIST Webbook
hfl	-260.00 ± 1.80	kJ/mol	NIST Webbook
hfus	14.82	kJ/mol	Joback Method
hvap	55.98	kJ/mol	NIST Webbook
hvap	55.98 ± 0.38	kJ/mol	NIST Webbook
hvap	56.00 ± 0.40	kJ/mol	NIST Webbook
log10ws	-0.96		Crippen Method
logp	1.481		Crippen Method
mcpol	100.800	ml/mol	McGowan Method
pc	4000.70	kPa	Joback Method
rinsol	1077.00		NIST Webbook
sl	274.40	J/mol×K	NIST Webbook
tb	426.86	K	Joback Method
tc	641.95	K	Joback Method
tf	270.84	K	Joback Method
tt	292.16 ± 0.02	K	NIST Webbook
vc	0.382	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.13	J/mol×K	426.86	Joback Method

cpg	194.95	J/mol×K	462.71	Joback Method
cpg	204.77	J/mol×K	498.56	Joback Method
cpg	213.69	J/mol×K	534.41	Joback Method
cpg	221.85	J/mol×K	570.26	Joback Method
cpg	229.35	J/mol×K	606.10	Joback Method
cpg	236.31	J/mol×K	641.95	Joback Method
cpl	218.70	J/mol×K	300.00	NIST Webbook
hfust	16.95	kJ/mol	292.20	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C78717&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C78717&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature

**vc:** Critical Volume

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