

# Phthalic acid, hexadecyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C32H54O4/c1-6-7-8-9-10-11-12-13-14-15-16-17-18-21-24-35-30(33)28-22-19-
InchiKey:	QNGDPTOSQFDMBS-UHFFFAOYSA-N
Formula:	C32H54O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)c1ccccc1C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	502.77

## Physical Properties

Property code	Value	Unit	Source
gf	-146.10	kJ/mol	Joback Method
hf	-982.38	kJ/mol	Joback Method
hfus	66.93	kJ/mol	Joback Method
hvap	106.39	kJ/mol	Joback Method
log10ws	-10.68		Crippen Method
logp	9.554		Crippen Method
mvol	452.860	ml/mol	McGowan Method
pc	672.90	kPa	Joback Method
rinpol	3381.00		NIST Webbook
rinpol	3381.00		NIST Webbook
tb	1112.13	K	Joback Method
tc	1379.62	K	Joback Method
tf	621.08	K	Joback Method
vc	1.750	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1625.54	J/molxK	1112.13	Joback Method
cpg	1706.04	J/molxK	1335.04	Joback Method
cpg	1693.31	J/molxK	1290.46	Joback Method
cpg	1679.06	J/molxK	1245.87	Joback Method
cpg	1663.13	J/molxK	1201.29	Joback Method
cpg	1645.34	J/molxK	1156.71	Joback Method
cpg	1717.42	J/molxK	1379.62	Joback Method
dvisc	0.0000059	Paxs	1112.13	Joback Method

dvisc	0.0000081	Paxs	1030.29	Joback Method
dvisc	0.0000116	Paxs	948.45	Joback Method
dvisc	0.0000178	Paxs	866.61	Joback Method
dvisc	0.0000299	Paxs	784.76	Joback Method
dvisc	0.0000568	Paxs	702.92	Joback Method
dvisc	0.0001278	Paxs	621.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377782&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377782&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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