

# tellurium

Inchi:	InChI=1S/Te
InchiKey:	PORWMNRCUJJQNO-UHFFFAOYSA-N
Formula:	Te
SMILES:	[Te]
Mol. weight [g/mol]:	127.60
CAS:	13494-80-9

## Physical Properties

Property code	Value	Unit	Source
tt	722.65 ± 0.30	K	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{\text{vap}}) = A + B/(T + C)$
Coeff. A	1.32707e+01
Coeff. B	-9.40463e+03
Coeff. C	-1.74210e+02
Temperature range (K), min.	775.15
Temperature range (K), max.	1261.15

## Sources

<b>Thermodynamic stability of Ca<sub>3</sub>TeO<sub>6</sub> determined by a solid electrolyte EMF Method:</b>	<a href="https://www.doi.org/10.1016/j.tca.2015.07.001">https://www.doi.org/10.1016/j.tca.2015.07.001</a> <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13494809&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13494809&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Thermodynamic properties of intermetallic PtTe determined by means Standard Gibbs energy of formation of tellurium dioxide measurement by a vaporization technique: Ni<sub>2</sub>Te<sub>3</sub>O<sub>8</sub>) in the Ni-Te-O system by transpiration thermogravimetry and Knudsen effusion mass spectrometry:</b>	<a href="https://www.doi.org/10.1016/j.jct.2015.09.007">https://www.doi.org/10.1016/j.jct.2015.09.007</a> <a href="https://www.doi.org/10.1016/j.tca.2013.08.006">https://www.doi.org/10.1016/j.tca.2013.08.006</a> <a href="https://www.doi.org/10.1016/j.tca.2014.11.024">https://www.doi.org/10.1016/j.tca.2014.11.024</a>

# Legend

**pvap:** Vapor pressure

**tt:** Triple Point Temperature

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