

# Pentafluorobenzaldehyde

<b>Other names:</b>	2,3,4,5,6-Pentafluorobenzaldehyde Benzaldehyde, pentafluoro- Benzaldehyde, 2,3,4,5,6-pentafluoro-
<b>Inchi:</b>	InChI=1S/C7HF5O/c8-3-2(1-13)4(9)6(11)7(12)5(3)10/h1H
<b>InchiKey:</b>	QJXCFMJTJYCLFG-UHFFFAOYSA-N
<b>Formula:</b>	C7HF5O
<b>SMILES:</b>	O=Cc1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	196.07
<b>CAS:</b>	653-37-2

## Physical Properties

Property code	Value	Unit	Source
ea	1.10 ± 0.11	eV	NIST Webbook
gf	-1001.25	kJ/mol	Joback Method
hf	-1074.76	kJ/mol	Joback Method
hfus	23.67	kJ/mol	Joback Method
hvap	39.40	kJ/mol	Joback Method
ie	10.35	eV	NIST Webbook
log10ws	-3.37		Crippen Method
logp	2.195		Crippen Method
mcvol	96.150	ml/mol	McGowan Method
pc	3138.51	kPa	Joback Method
rinpola	943.00		NIST Webbook
rinpola	943.00		NIST Webbook
tb	438.00 ± 1.00	K	NIST Webbook
tb	438.20	K	NIST Webbook
tb	438.00 ± 1.00	K	NIST Webbook
tc	629.33	K	Joback Method
tf	302.62	K	Joback Method
vc	0.426	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	197.90	J/mol×K	456.15	Joback Method
cpg	203.81	J/mol×K	485.01	Joback Method
cpg	209.51	J/mol×K	513.88	Joback Method
cpg	214.98	J/mol×K	542.74	Joback Method
cpg	220.25	J/mol×K	571.61	Joback Method
cpg	225.29	J/mol×K	600.47	Joback Method
cpg	230.12	J/mol×K	629.33	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C653372&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C653372&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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