

# 1-Fluorooctane

<b>Other names:</b>	Octane, 1-fluoro- Octyl fluoride n-Octyl fluoride
<b>Inchi:</b>	InChI=1S/C8H17F/c1-2-3-4-5-6-7-8-9/h2-8H2,1H3
<b>InchiKey:</b>	DHIVLKMKGKIZOHF-UHFFFAOYSA-N
<b>Formula:</b>	C8H17F
<b>SMILES:</b>	CCCCCCCCF
<b>Mol. weight [g/mol]:</b>	132.22
<b>CAS:</b>	463-11-6

## Physical Properties

Property code	Value	Unit	Source
chl	-5318.00	kJ/mol	NIST Webbook
gf	-178.33	kJ/mol	Joback Method
hf	-404.56	kJ/mol	Joback Method
hfus	19.56	kJ/mol	Joback Method
hvap	49.68	kJ/mol	NIST Webbook
log10ws	-3.02		Crippen Method
logp	3.316		Crippen Method
mcvol	125.350	ml/mol	McGowan Method
pc	2424.27	kPa	Joback Method
rinpol	857.90		NIST Webbook
rinpol	859.60		NIST Webbook
tb	415.50	K	NIST Webbook
tb	415.70	K	NIST Webbook
tc	536.73	K	Joback Method
tf	180.51	K	Joback Method
vc	0.501	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	240.42	J/mol×K	381.71	Joback Method
cpg	252.82	J/mol×K	407.55	Joback Method

cpg	264.79	J/mol×K	433.38	Joback Method
cpg	276.35	J/mol×K	459.22	Joback Method
cpg	287.50	J/mol×K	485.05	Joback Method
cpg	298.25	J/mol×K	510.89	Joback Method
cpg	308.62	J/mol×K	536.73	Joback Method
hvapt	44.10	kJ/mol	376.50	NIST Webbook
hvapt	40.43	kJ/mol	415.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45988e+01
Coeff. B	-3.58804e+03
Coeff. C	-5.66450e+01
Temperature range (K), min.	307.36
Temperature range (K), max.	442.98

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C463116&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C463116&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>h<sub>vapt</sub>:</b>	Enthalpy of vaporization at a given temperature
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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