

# Benzene, 1-fluoro-2-(trifluoromethyl)-

<b>Other names:</b>	o-Fluorobenzotrifluoride 2-Fluorobenzotrifluoride o,«alpha»,«alpha»,«alpha»-Tetrafluorotoluene Toluene, «alpha»,«alpha»,«alpha»,o-tetrafluoro- Toluene, o,«alpha»,«alpha»,«alpha»-tetrafluoro- 1-Fluoro-2-(trifluoromethyl)benzene o,alpha,alpha,alpha-Tetrafluorotoluene NSC 10314 «alpha»,«alpha»,«alpha»,2-tetrafluorotoluene
<b>Inchi:</b>	InChI=1S/C7H4F4/c8-6-4-2-1-3-5(6)7(9,10)11/h1-4H
<b>InchiKey:</b>	BGVGHYOIWIALFF-UHFFFAOYSA-N
<b>Formula:</b>	C7H4F4
<b>SMILES:</b>	Fc1ccccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	164.10
<b>CAS:</b>	392-85-8

## Physical Properties

Property code	Value	Unit	Source
gf	-665.56	kJ/mol	Joback Method
hf	-755.94	kJ/mol	Joback Method
hfus	12.44	kJ/mol	Joback Method
hvap	38.10	kJ/mol	NIST Webbook
log10ws	-2.91		Crippen Method
logp	2.845		Crippen Method
mcvol	92.810	ml/mol	McGowan Method
pc	3322.01	kPa	Joback Method
tb	387.50 ± 0.50	K	NIST Webbook
tc	566.36	K	Joback Method
tf	212.37	K	Joback Method
vc	0.381	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	174.38	J/mol×K	385.07	Joback Method
cpg	184.55	J/mol×K	415.28	Joback Method
cpg	194.08	J/mol×K	445.50	Joback Method
cpg	203.00	J/mol×K	475.71	Joback Method
cpg	211.34	J/mol×K	505.93	Joback Method
cpg	219.11	J/mol×K	536.14	Joback Method
cpg	226.36	J/mol×K	566.36	Joback Method
hfust	10.70	kJ/mol	222.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	387.70	K	100.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C392858&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C392858&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tbrp:** Boiling point at reduced pressure  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/54-614-0/Benzene-1-fluoro-2-trifluoromethyl.pdf>

Generated by Cheméo on 2024-05-15 12:00:36.060644879 +0000 UTC m=+18063684.981222190.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.