

# Butane, 2,2,3,3-tetramethyl-

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>3</sub> CC(CH <sub>3</sub> ) <sub>3</sub> 2,2,3,3-Tetramethylbutane Ethane, hexamethyl- HEXAMETHYLETHANE TETRAMETHYLBUTANE
<b>Inchi:</b>	InChI=1S/C8H18/c1-7(2,3)8(4,5)6/h1-6H3
<b>InchiKey:</b>	OMMLUKLXGSRPHK-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>8</sub> H <sub>18</sub>
<b>SMILES:</b>	CC(C)(C)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	114.23
<b>CAS:</b>	594-82-1

## Physical Properties

Property code	Value	Unit	Source
af	0.2510		KDB
chs	-5451.50 ± 1.10	kJ/mol	NIST Webbook
chs	-5451.90 ± 1.80	kJ/mol	NIST Webbook
gf	22.16	kJ/mol	Joback Method
hf	-226.20	kJ/mol	NIST Webbook
hf	-225.90 ± 1.90	kJ/mol	NIST Webbook
hfs	-268.80 ± 1.90	kJ/mol	NIST Webbook
hfs	-269.10 ± 1.20	kJ/mol	NIST Webbook
hfus	1.65	kJ/mol	Joback Method
hsub	42.90	kJ/mol	NIST Webbook
hsub	43.37 ± 0.21	kJ/mol	NIST Webbook
hsub	43.40 ± 0.20	kJ/mol	NIST Webbook
hsub	42.90 ± 0.90	kJ/mol	NIST Webbook
hvap	42.91	kJ/mol	NIST Webbook
hvap	42.94	kJ/mol	NIST Webbook
ie	9.80	eV	NIST Webbook
ie	10.20 ± 0.05	eV	NIST Webbook
ie	10.37	eV	NIST Webbook
ie	9.80	eV	NIST Webbook
ie	9.79	eV	NIST Webbook
log10ws	-2.69		Crippen Method
logp	3.079		Crippen Method
mcvol	123.580	ml/mol	McGowan Method

pc	2870.00	kPa	KDB
rinpol	720.00		NIST Webbook
rinpol	731.00		NIST Webbook
rinpol	773.00		NIST Webbook
rinpol	736.00		NIST Webbook
rinpol	733.28		NIST Webbook
rinpol	736.21		NIST Webbook
rinpol	739.23		NIST Webbook
rinpol	727.21		NIST Webbook
rinpol	730.14		NIST Webbook
rinpol	723.90		NIST Webbook
rinpol	722.00		NIST Webbook
rinpol	724.00		NIST Webbook
rinpol	726.00		NIST Webbook
rinpol	728.00		NIST Webbook
rinpol	720.00		NIST Webbook
rinpol	723.70		NIST Webbook
rinpol	724.00		NIST Webbook
rinpol	720.00		NIST Webbook
rinpol	726.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	735.00		NIST Webbook
rinpol	721.00		NIST Webbook
rinpol	726.00		NIST Webbook
rinpol	731.00		NIST Webbook
rinpol	723.00		NIST Webbook
rinpol	728.00		NIST Webbook
rinpol	731.00		NIST Webbook
rinpol	729.00		NIST Webbook
rinpol	720.00		NIST Webbook
rinpol	716.00		NIST Webbook
rinpol	710.35		NIST Webbook
rinpol	712.98		NIST Webbook
rinpol	714.69		NIST Webbook
rinpol	712.25		NIST Webbook
rinpol	730.20		NIST Webbook
rinpol	716.46		NIST Webbook
rinpol	726.10		NIST Webbook
rinpol	719.00		NIST Webbook
rinpol	735.00		NIST Webbook
rinpol	676.00		NIST Webbook
rinpol	729.00		NIST Webbook
rinpol	724.00		NIST Webbook
rinpol	714.80		NIST Webbook

rinpol	733.00		NIST Webbook
rinpol	726.00		NIST Webbook
sg	389.30 ± 1.30	J/molxK	NIST Webbook
ss	256.90	J/molxK	NIST Webbook
ss	273.76	J/molxK	NIST Webbook
tb	379.60	K	KDB
tc	567.80	K	KDB
tf	373.80	K	KDB
tt	373.97 ± 0.07	K	NIST Webbook
vc	0.461	m <sup>3</sup> /kmol	KDB
zc	0.2802530		KDB
zra	0.27		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	294.16	J/molxK	500.01	Joback Method
cpg	307.15	J/molxK	531.02	Joback Method
cpg	233.87	J/molxK	375.98	Joback Method
cpg	250.27	J/molxK	406.99	Joback Method
cpg	265.77	J/molxK	438.00	Joback Method
cpg	280.38	J/molxK	469.01	Joback Method
cpg	319.38	J/molxK	562.03	Joback Method
cps	239.62	J/molxK	301.60	NIST Webbook
cps	232.20	J/molxK	295.40	NIST Webbook
dvisc	0.0007345	Paxs	312.24	Joback Method
dvisc	0.0004645	Paxs	344.11	Joback Method
dvisc	0.0003175	Paxs	375.98	Joback Method
dvisc	0.0223115	Paxs	184.76	Joback Method
dvisc	0.0065211	Paxs	216.63	Joback Method
dvisc	0.0026130	Paxs	248.50	Joback Method
dvisc	0.0012890	Paxs	280.37	Joback Method
hfust	7.54	kJ/mol	373.90	NIST Webbook
hfust	7.54	kJ/mol	373.90	NIST Webbook
hfust	2.00	kJ/mol	152.50	NIST Webbook
hsubt	43.60	kJ/mol	331.50	NIST Webbook
hsubt	56.20	kJ/mol	271.00	NIST Webbook
hvapt	333.00	kJ/mol	383.50	NIST Webbook
hvapt	31.42	kJ/mol	379.60	KDB
sfust	20.16	J/molxK	373.90	NIST Webbook
sfust	13.11	J/molxK	152.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	379.70	K	102.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.31777e+01
Coeff. B	-2.51762e+03
Coeff. C	-8.53050e+01
Temperature range (K), min.	280.62
Temperature range (K), max.	405.36

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.32058e+02
Coeff. B	-9.96316e+03
Coeff. C	-1.72776e+01
Coeff. D	9.39147e-06
Temperature range (K), min.	270.00
Temperature range (K), max.	567.80

## Sources

<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol63.mol">https://www.cheric.org/files/research/kdb/mol/mol63.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C594821&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C594821&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=63">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=63</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>af:</b>	Acentric Factor
<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sg:</b>	Molar entropy at standard conditions
<b>srf:</b>	Surface Tension
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility
<b>zra:</b>	Rackett Parameter

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