

# Benzene, 1,1'-(2,2-dichloroethylidene)bis-

<b>Inchi:</b>	InChI=1S/C14H12Cl2/c15-14(16)13(11-7-3-1-4-8-11)12-9-5-2-6-10-12/h1-10,13-14H
<b>InchiKey:</b>	FVMUDWLRSAABPN-UHFFFAOYSA-N
<b>Formula:</b>	C14H12Cl2
<b>SMILES:</b>	C1C(Cl)C(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	251.15
<b>CAS:</b>	2387-16-8

## Physical Properties

Property code	Value	Unit	Source
gf	263.08	kJ/mol	Joback Method
hf	98.73	kJ/mol	Joback Method
hfus	21.45	kJ/mol	Joback Method
hvap	59.30	kJ/mol	Joback Method
log10ws	-4.87		Crippen Method
logp	4.622		Crippen Method
mcvol	185.080	ml/mol	McGowan Method
pc	2608.40	kPa	Joback Method
rinpol	1854.00		NIST Webbook
rinpol	1854.00		NIST Webbook
tb	647.06	K	Joback Method
tc	904.20	K	Joback Method
tf	330.22	K	Joback Method
vc	0.690	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	427.88	J/mol×K	647.06	Joback Method
cpg	493.65	J/mol×K	861.34	Joback Method
cpg	482.90	J/mol×K	818.48	Joback Method
cpg	471.06	J/mol×K	775.63	Joback Method
cpg	458.01	J/mol×K	732.77	Joback Method
cpg	443.65	J/mol×K	689.92	Joback Method
cpg	503.41	J/mol×K	904.20	Joback Method

dvisc	0.0001421	Paxs	647.06	Joback Method
dvisc	0.0001899	Paxs	594.25	Joback Method
dvisc	0.0002687	Paxs	541.45	Joback Method
dvisc	0.0004097	Paxs	488.64	Joback Method
dvisc	0.0006920	Paxs	435.83	Joback Method
dvisc	0.0013505	Paxs	383.03	Joback Method
dvisc	0.0032642	Paxs	330.22	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2387168&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2387168&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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<https://www.chemeo.com/cid/53-078-7/Benzene-1-1-2-2-dichloroethylidene-bis.pdf>

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