

# lithium iodide

Inchi:	InChI=1S/HI.Li/h1H;/q;+1/p-1
InchiKey:	HSZCZNFUXUDYRKD-UHFFFAOYSA-M
Formula:	LI
SMILES:	[I-].[Li+]
Mol. weight [g/mol]:	133.84
CAS:	10377-51-2

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.75421e+01
Coeff. B	-1.71770e+04
Coeff. C	-1.17900e+02
Temperature range (K), min.	996.15
Temperature range (K), max.	1447.00

## Sources

Electrical Conductivity of Lithium Chloride, Lithium Bromide, and Lithium Iodide in Aqueous Solutions at Various Temperatures and Concentrations	<a href="https://www.doi.org/10.1021/acs.jced.9b00405">https://www.doi.org/10.1021/acs.jced.9b00405</a>
Conductivity of Aqueous Alkali Halide Solutions	<a href="https://www.doi.org/10.1016/j.jct.2009.03.005">https://www.doi.org/10.1016/j.jct.2009.03.005</a>
Density of Aqueous Alkali Halide Salt Solutions by Experiment and Molecular Simulation	<a href="https://www.doi.org/10.1021/je5009944">https://www.doi.org/10.1021/je5009944</a>
Interaction of lithium salts in binary mixtures of acetonitrile and diethyl ether	<a href="https://www.doi.org/10.1016/j.fluid.2013.08.022">https://www.doi.org/10.1016/j.fluid.2013.08.022</a>
Solvent interactions of some chemical halides of common cations with organic solvents	<a href="https://www.doi.org/10.1021/je900656c">https://www.doi.org/10.1021/je900656c</a>
Chemistry of lithium halides in 1,2-dioxolane and its binary mixtures with diethyl ether	<a href="https://www.doi.org/10.1016/j.tca.2012.08.009">https://www.doi.org/10.1016/j.tca.2012.08.009</a>
Viscosity and refractive index of lithium halide solutions	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Thermodynamic, volumetric, and viscometric dependence of the density of aqueous alkali halide salt solutions	<a href="https://www.doi.org/10.1021/je500420g">https://www.doi.org/10.1021/je500420g</a>
Densities and apparent molar volumes of aqueous LiI solutions at temperatures from 298.15 K to 323.15 K and at pressures from 0.1 MPa to 40 MPa	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10377512&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10377512&amp;Units=SI</a>
Density of ethanolic alkali halide salt solutions by experiment and molecular simulation	<a href="https://www.doi.org/10.1016/j.jct.2004.06.001">https://www.doi.org/10.1016/j.jct.2004.06.001</a>
Density of aqueous LiI solutions at 293.15 K and 0.1 MPa	<a href="https://www.doi.org/10.1016/j.jct.2013.08.018">https://www.doi.org/10.1016/j.jct.2013.08.018</a>
Density of aqueous LiI solutions at 293.15 K and 0.1 MPa	<a href="https://www.doi.org/10.1016/j.tca.2005.08.036">https://www.doi.org/10.1016/j.tca.2005.08.036</a>
Density of ethanolic alkali halide salt solutions by experiment and molecular simulation	<a href="https://www.doi.org/10.1016/j.fluid.2015.08.005">https://www.doi.org/10.1016/j.fluid.2015.08.005</a>

# Legend

**pvap:** Vapor pressure

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