

# (CH<sub>3</sub>)<sub>2</sub>N-C(CH<sub>3</sub>)=NC<sub>2</sub>H<sub>5</sub>

<b>Inchi:</b>	InChI=1S/C6H14N2/c1-5-7-6(2)8(3)4/h5H2,1-4H3
<b>InchiKey:</b>	ISBGMIISPIKHNE-UHFFFAOYSA-N
<b>Formula:</b>	C6H14N2
<b>SMILES:</b>	CCN=C(C)N(C)C
<b>Mol. weight [g/mol]:</b>	114.19
<b>CAS:</b>	151328-40-4

## Physical Properties

Property code	Value	Unit	Source
affp	1029.10	kJ/mol	NIST Webbook
basg	996.70	kJ/mol	NIST Webbook
hf	-27.21	kJ/mol	Joback Method
hvap	34.39	kJ/mol	Joback Method
log10ws	-0.56		Crippen Method
logp	0.986		Crippen Method
mcvol	111.060	ml/mol	McGowan Method
pc	2781.78	kPa	Joback Method
tb	425.68	K	Joback Method
tc	615.03	K	Joback Method

## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C151328404&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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