

# 1-ethyl-cis-2-hexyl-cyclopropane

<b>Inchi:</b>	InChI=1S/C11H22/c1-3-5-6-7-8-11-9-10(11)4-2/h10-11H,3-9H2,1-2H3/t10-,11+/m1/s1
<b>InchiKey:</b>	YPODGGYYHXRDLB-MNOVXSKESA-N
<b>Formula:</b>	C11H22
<b>SMILES:</b>	CCCCCCC1CC1CC
<b>Mol. weight [g/mol]:</b>	154.29

## Physical Properties

Property code	Value	Unit	Source
gf	94.78	kJ/mol	Joback Method
hf	-217.91	kJ/mol	Joback Method
hfus	23.45	kJ/mol	Joback Method
hvap	39.68	kJ/mol	Joback Method
log10ws	-3.84		Crippen Method
logp	4.003		Crippen Method
mcvol	154.990	ml/mol	McGowan Method
pc	2110.00	kPa	Joback Method
rinpol	1088.80		NIST Webbook
rinpol	1087.20		NIST Webbook
rinpol	1087.20		NIST Webbook
tb	453.15	K	Joback Method
tc	626.88	K	Joback Method
tf	227.43	K	Joback Method
vc	0.608	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	346.16	J/molxK	453.15	Joback Method
cpg	426.54	J/molxK	597.93	Joback Method
cpg	411.90	J/molxK	568.97	Joback Method
cpg	396.58	J/molxK	540.02	Joback Method
cpg	380.53	J/molxK	511.06	Joback Method
cpg	363.74	J/molxK	482.11	Joback Method
cpg	440.50	J/molxK	626.88	Joback Method

dvisc	0.0004076	Paxs	453.15	Joback Method
dvisc	0.0004579	Paxs	415.53	Joback Method
dvisc	0.0005265	Paxs	377.91	Joback Method
dvisc	0.0006243	Paxs	340.29	Joback Method
dvisc	0.0007724	Paxs	302.67	Joback Method
dvisc	0.0010151	Paxs	265.05	Joback Method
dvisc	0.0014603	Paxs	227.43	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R485997&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R485997&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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