

# (4-(2'-Propenyl)phenyl)phenyl ether

<b>Inchi:</b>	InChI=1S/C15H14O2/c1-2-12-16-13-8-10-15(11-9-13)17-14-6-4-3-5-7-14/h2-11H,1,12H2
<b>InchiKey:</b>	NAYOLUZVOCYYFZ-UHFFFAOYSA-N
<b>Formula:</b>	C15H14O2
<b>SMILES:</b>	C=CCOc1ccc(Oc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	226.27

## Physical Properties

Property code	Value	Unit	Source
gf	168.45	kJ/mol	Joback Method
hf	-30.35	kJ/mol	Joback Method
hfus	23.39	kJ/mol	Joback Method
hvap	58.35	kJ/mol	Joback Method
log10ws	-3.92		Crippen Method
logp	4.044		Crippen Method
mvol	182.130	ml/mol	McGowan Method
pc	2492.52	kPa	Joback Method
rinpol	1744.00		NIST Webbook
rinpol	1744.00		NIST Webbook
tb	642.46	K	Joback Method
tc	876.21	K	Joback Method
tf	366.87	K	Joback Method
vc	0.676	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	451.98	J/molxK	642.46	Joback Method
cpg	468.31	J/molxK	681.42	Joback Method
cpg	483.48	J/molxK	720.38	Joback Method
cpg	497.53	J/molxK	759.33	Joback Method
cpg	510.49	J/molxK	798.29	Joback Method
cpg	522.40	J/molxK	837.25	Joback Method
cpg	533.29	J/molxK	876.21	Joback Method
dvisc	0.0010202	Paxs	366.87	Joback Method

dvisc	0.0005703	Paxs	412.80	Joback Method
dvisc	0.0003582	Paxs	458.73	Joback Method
dvisc	0.0002449	Paxs	504.67	Joback Method
dvisc	0.0001783	Paxs	550.60	Joback Method
dvisc	0.0001364	Paxs	596.53	Joback Method
dvisc	0.0001084	Paxs	642.46	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R569134&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R569134&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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