

# 3-Hydroxy-2,2,4-trimethyl-3-pentenoic acid «beta»-lactone

<b>Other names:</b>	Dimethylketene-«beta»-lactone dimer 2,2,4-Trimethyl-3-hydroxy-3-pentenoic acid «beta»-lactone 3-Hydroxy-2,4,4-trimethyl-3-pentenoic acid «beta»-lactone 2-Oxetanone, 3,3-dimethyl-4-(1-methylethylidene)- Dimethylketene dimer 2-Oxetanone, 4-isopropylidene-3,3-dimethyl- 4-Isopropylidene-3,3-dimethyl-2-oxetanone 3,3-Dimethyl-4-(1-methylethylidene)-2-oxetanone 2,2,4-Trimethylpent-3-en-3-olide NSC 74312 2,2-dimethyl-3-isopropyliden-«beta»-propiolactone
<b>Inchi:</b>	InChI=1S/C8H12O2/c1-5(2)6-8(3,4)7(9)10-6/h1-4H3
<b>InchiKey:</b>	OKAAPXFKYOTAEN-UHFFFAOYSA-N
<b>Formula:</b>	C8H12O2
<b>SMILES:</b>	<chem>CC(C)=C1OC(=O)C1(C)C</chem>
<b>Mol. weight [g/mol]:</b>	140.18
<b>CAS:</b>	3173-79-3

## Physical Properties

Property code	Value	Unit	Source
gf	-112.16	kJ/mol	Joback Method
hf	-330.03	kJ/mol	Joback Method
hfus	12.71	kJ/mol	Joback Method
hvap	41.96	kJ/mol	Joback Method
log10ws	-2.04		Crippen Method
logp	1.863		Crippen Method
mcvol	115.860	ml/mol	McGowan Method
pc	3356.75	kPa	Joback Method
tb	443.20	K	NIST Webbook
tc	718.55	K	Joback Method
tf	309.43	K	Joback Method
vc	0.443	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	261.54	J/molxK	494.98	Joback Method
cpg	274.88	J/molxK	532.24	Joback Method
cpg	287.39	J/molxK	569.50	Joback Method
cpg	299.16	J/molxK	606.77	Joback Method
cpg	310.32	J/molxK	644.03	Joback Method
cpg	320.94	J/molxK	681.29	Joback Method
cpg	331.15	J/molxK	718.55	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3173793&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3173793&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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