

# Tetrabromo-1,4-benzoquinone

<b>Other names:</b>	p-Bromanil Tetrabromo-p-benzoquinone Bromanil 2,5-Cyclohexadiene-1,4-dione, 2,3,5,6-tetrabromo- p-Benzoquinone, 2,3,5,6-tetrabromo- Tetrabromoquinone 2,3,5,6-Tetrabromo-p-benzoquinone 2,3,5,6-Tetrabromo-1,4-benzoquinone p-Bromoanil NSC 36927 p-Benzoquinone, tetrabromo- p-Quinone, tetrabromo- Tetrabromobenzoquinone Tetrabromo-p-quinone
<b>Inchi:</b>	InChI=1S/C6Br4O2/c7-1-2(8)6(12)4(10)3(9)5(1)11
<b>InchiKey:</b>	LWHDQPLUIFIFFT-UHFFFAOYSA-N
<b>Formula:</b>	C6Br4O2
<b>SMILES:</b>	O=C1C(Br)=C(Br)C(=O)C(Br)=C1Br
<b>Mol. weight [g/mol]:</b>	423.68
<b>CAS:</b>	488-48-2

## Physical Properties

Property code	Value	Unit	Source
gf	-134.70	kJ/mol	Joback Method
hf	-192.91	kJ/mol	Joback Method
hfus	23.11	kJ/mol	Joback Method
hvap	67.15	kJ/mol	Joback Method
ie	9.59	eV	NIST Webbook
log10ws	-4.20		Crippen Method
logp	3.141		Crippen Method
mcvol	149.080	ml/mol	McGowan Method
pc	6967.65	kPa	Joback Method
tb	779.42	K	Joback Method
tc	1086.46	K	Joback Method
tf	596.24	K	Joback Method
vc	0.539	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.69	J/mol×K	779.42	Joback Method
cpg	252.80	J/mol×K	830.59	Joback Method
cpg	258.12	J/mol×K	881.77	Joback Method
cpg	262.60	J/mol×K	932.94	Joback Method
cpg	266.16	J/mol×K	984.12	Joback Method
cpg	268.74	J/mol×K	1035.29	Joback Method
cpg	270.27	J/mol×K	1086.46	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C488482&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C488482&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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