

# Diethylamine, 2,2'-dichloro-

<b>Other names:</b>	N,N-Bis-(«beta»-chloroethyl)-amine Bis-«beta»-chloroethylamine NH-lost Nor-nitrogen mustard NSC-10873 Ethanamine, 2-chloro-N-(2-chloroethyl)- 2,2'-Dichlorodiethylamine Bis(beta-chloroethyl)amine Bis(2-chloroethyl)amine Bis(chloroethyl)amine Di(2-chloroethyl)amine N,N-Bis(2-chloroethane)amine Nor-HN2 Nor-mechlorethamine Nor-mustard N,N-Bis(2-chloroethyl)amine Nor-N-mustard Nitrogen mustard
<b>Inchi:</b>	InChI=1S/C4H9Cl2N/c5-1-3-7-4-2-6/h7H,1-4H2
<b>InchiKey:</b>	TXFLGZOGNOOEFZ-UHFFFAOYSA-N
<b>Formula:</b>	C4H9Cl2N
<b>SMILES:</b>	C1CCNCCC1
<b>Mol. weight [g/mol]:</b>	142.03
<b>CAS:</b>	334-22-5

## Physical Properties

Property code	Value	Unit	Source
gf	48.33	kJ/mol	Joback Method
hf	-103.90	kJ/mol	Joback Method
hfus	19.61	kJ/mol	Joback Method
hvap	39.70	kJ/mol	Joback Method
log10ws	-0.99		Crippen Method
logp	1.054		Crippen Method
mcvol	101.680	ml/mol	McGowan Method
pc	3615.89	kPa	Joback Method
rinpol	1087.00		NIST Webbook
tb	415.95	K	Joback Method

tc	603.01	K	Joback Method
tf	247.34	K	Joback Method
vc	0.393	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	176.49	J/mol×K	415.95	Joback Method
cpg	184.77	J/mol×K	447.13	Joback Method
cpg	192.66	J/mol×K	478.30	Joback Method
cpg	200.19	J/mol×K	509.48	Joback Method
cpg	207.35	J/mol×K	540.66	Joback Method
cpg	214.18	J/mol×K	571.84	Joback Method
cpg	220.66	J/mol×K	603.01	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C334225&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C334225&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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