# **Cycloheptanol**

Other names: SUBEROL

InChl=1S/C7H14O/c8-7-5-3-1-2-4-6-7/h7-8H,1-6H2

InchiKey: QCRFMSUKWRQZEM-UHFFFAOYSA-N

Formula: C7H14O

SMILES: OC1CCCCC1

Mol. weight [g/mol]: 114.19 CAS: 502-41-0

### **Physical Properties**

Property code	Value	Unit	Source
chl	-4362.00 ± 1.00	kJ/mol	NIST Webbook
chl	-4429.20	kJ/mol	NIST Webbook
gf	-116.41	kJ/mol	Joback Method
hf	-291.88	kJ/mol	Joback Method
hfus	7.71	kJ/mol	Joback Method
hvap	48.46	kJ/mol	Joback Method
log10ws	-0.88		Aqueous Solubility Prediction Method
log10ws	-0.88		Estimated Solubility Method
logp	1.701		Crippen Method
mcvol	104.500	ml/mol	McGowan Method
рс	4067.32	kPa	Joback Method
rinpol	1040.20		NIST Webbook
rinpol	1022.00		NIST Webbook
rinpol	1028.00		NIST Webbook
ripol	1547.00		NIST Webbook
ripol	1566.00		NIST Webbook
ripol	1566.00		NIST Webbook
ripol	1566.00		NIST Webbook
sl	241.64	J/mol×K	NIST Webbook
tb	458.20	K	NIST Webbook
tc	675.73	K	Joback Method
tf	233.33	K	Joback Method
tt	280.30 ± 0.02	K	NIST Webbook
VC	0.371	m3/kmol	Joback Method

## **Temperature Dependent Properties**

Property code	Value	Unit	Temperature [K]	Source	
cpg	230.00	J/mol×K	475.56	Joback Method	
cpg	308.33	J/mol×K	675.73	Joback Method	
cpg	244.83	J/mol×K	508.92	Joback Method	
cpg	258.94	J/mol×K	542.28	Joback Method	
cpg	272.33	J/mol×K	575.64	Joback Method	
cpg	285.01	J/mol×K	609.00	Joback Method	
cpg	297.01	J/mol×K	642.37	Joback Method	
cpl	250.22	J/mol×K	298.15	NIST Webbook	
cpl	244.30	J/mol×K	298.00	NIST Webbook	
cps	198.10	J/mol×K	264.00	Heat capacities of selected cycloalcohols	
cps	199.80	J/mol×K	266.00	Heat capacities of selected cycloalcohols	
cps	199.80	J/mol×K	266.00	Heat capacities of selected cycloalcohols	
cps	198.00	J/mol×K	264.00	Heat capacities of selected cycloalcohols	
cps	196.50	J/mol×K	262.00	Heat capacities of selected cycloalcohols	
cps	196.40	J/mol×K	262.00	Heat capacities of selected cycloalcohols	
cps	195.30	J/mol×K	260.59	Heat capacities of selected cycloalcohols	
cps	201.80	J/mol×K	268.00	Heat capacities of selected cycloalcohols	
cps	201.70	J/mol×K	268.00	Heat capacities of selected cycloalcohols	
cps	204.00	J/mol×K	270.00	Heat capacities of selected cycloalcohols	
cps	204.00	J/mol×K	270.00	Heat capacities of selected cycloalcohols	
cps	205.20	J/mol×K	271.34	Heat capacities of selected cycloalcohols	

cps	205.40	J/mol×K	271.43	Heat capacities of selected cycloalcohols	
cps	195.40	J/mol×K	260.73	Heat capacities of selected cycloalcohols	
dvisc	0.0044230	Paxs	314.07	Joback Method	
dvisc	0.0006163	Paxs	394.82	Joback Method	
dvisc	0.0003026	Paxs	435.19	Joback Method	
dvisc	0.0014757	Paxs	354.44	Joback Method	
dvisc	0.1241815	Paxs	233.33	Joback Method	
dvisc	0.0183260	Paxs	273.70	Joback Method	
dvisc	0.0001677	Paxs	475.56	Joback Method	
hfust	2.93	kJ/mol	172.20	NIST Webbook	
hfust	0.88	kJ/mol	258.40	NIST Webbook	
hfust	1.60	kJ/mol	280.30	NIST Webbook	
hfust	0.55	kJ/mol	227.30	NIST Webbook	
hfust	1.51	kJ/mol	278.30	NIST Webbook	
hfust	1.60	kJ/mol	280.30	NIST Webbook	
hvapt	64.70	kJ/mol	303.50	NIST Webbook	
hvapt	67.40	kJ/mol	302.50	NIST Webbook	
pvap	7.50e-03	kPa	284.67	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.07	kPa	308.15	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.08	kPa	310.11	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	

pvap	0.16	kPa	319.51	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.06	kPa	307.00	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.04	kPa	303.37	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.03	kPa	298.44	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.03	kPa	298.28	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	

pvap	0.03	kPa	298.23	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.02	kPa	293.86	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.12	kPa	315.14	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	7.10e-03	kPa	284.35	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.02	kPa	293.72	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	

pvap	0.02	kPa	292.50	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.01	kPa	288.22	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.02	kPa	293.73	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
pvap	0.18	kPa	320.88	Vapor Pressure and Its Temperature Dependence of 28 Organic Compounds: Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols	
sfust	5.72	J/mol×K	280.30	NIST Webbook	
sfust	16.98	J/mol×K	172.20	NIST Webbook	
sfust	2.44	J/mol×K	227.30	NIST Webbook	
sfust	3.39	J/mol×K	258.40	NIST Webbook	
srf	0.02	N/m	323.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	
srf	0.03	N/m	298.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	

srf	0.03	N/m	293.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	
srf	0.03	N/m	318.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	
srf	0.03	N/m	313.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	
srf	0.03	N/m	308.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	
srf	0.03	N/m	303.15	Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol	

#### **Correlations**

Information Value

Property code	pvap	
Equation	$ln(Pvp) = A + B/T + C*ln(T) + D*T^2$	
Coeff. A	-2.66915e+01	
Coeff. B	-5.85520e+03	
Coeff. C	7.60167e+00	
Coeff. D	-6.45886e-06	
Temperature range (K), min.	284.15	
Temperature range (K), max.	323.15	

### **Sources**

KDB: https://www.cheric.org/files/research/kdb/mol/mol900.mol

Joback Method: https://en.wikipedia.org/wiki/Joback\_method

Surface Tensions of Dilute Solutions of Cycloheptanol in Ethylene Glycol: Vapor Pressure and its Temperature Dependence of 28 Organic Wendwards Cyclic Amines, Cyclic Ethers, and Cyclic and Open Chain Secondary Alcohols:

https://www.doi.org/10.1021/je050055m h

https://www.doi.org/10.1021/acs.jced.6b00576

http://link.springer.com/article/10.1007/BF02311772

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

KDB Vapor Pressure Data: https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=900

Estimated Solubility Method: http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\_file/ci034243xsi20040112\_053635.txt

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C502410&Units=SI

Heat capacities of selected

cycloalcohols:

https://www.doi.org/10.1016/j.tca.2014.10.002

#### Legend

**chl:** Standard liquid enthalpy of combustion

**cpg:** Ideal gas heat capacity

cpl: Liquid phase heat capacitycps: Solid phase heat capacity

**dvisc:** Dynamic viscosity

gf: Standard Gibbs free energy of formation
hf: Enthalpy of formation at standard conditions
hfus: Enthalpy of fusion at a given temperature

hvap: Enthalpy of vaporization at standard conditions hvapt: Enthalpy of vaporization at a given temperature

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressurepvap: Vapor pressure

rinpol: Non-polar retention indices

ripol: Polar retention indices

**sfust:** Entropy of fusion at a given temperature

**sl:** Liquid phase molar entropy at standard conditions

**srf:** Surface Tension

**tb:** Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) pointtt: Triple Point Temperature

vc: Critical Volume

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