

# Diethylmalonic acid, 8-chlorooctyl isobutyl ester

Inchi:	InChI=1S/C19H35ClO4/c1-5-19(6-2,18(22)24-15-16(3)4)17(21)23-14-12-10-8-7-9-11-13
InchiKey:	LMXJIWNPDHVCFW-UHFFFAOYSA-N
Formula:	C19H35ClO4
SMILES:	CCC(CC)(C(=O)OCCCCCCCCCl)C(=O)OCC(C)C
Mol. weight [g/mol]:	362.93

## Physical Properties

Property code	Value	Unit	Source
gf	-370.27	kJ/mol	Joback Method
hf	-954.86	kJ/mol	Joback Method
hfus	43.80	kJ/mol	Joback Method
hvap	78.90	kJ/mol	Joback Method
log10ws	-5.17		Crippen Method
logp	5.115		Crippen Method
mcvol	305.690	ml/mol	McGowan Method
pc	1141.34	kPa	Joback Method
rinqol	2262.00		NIST Webbook
tb	820.46	K	Joback Method
tc	1011.44	K	Joback Method
tf	465.55	K	Joback Method
vc	1.179	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	942.07	J/molxK	820.46	Joback Method
cpg	959.09	J/molxK	852.29	Joback Method
cpg	975.06	J/molxK	884.12	Joback Method
cpg	990.02	J/molxK	915.95	Joback Method
cpg	1003.99	J/molxK	947.78	Joback Method
cpg	1017.01	J/molxK	979.61	Joback Method
cpg	1029.12	J/molxK	1011.44	Joback Method
dvisc	0.0008193	Paxs	465.55	Joback Method
dvisc	0.0003697	Paxs	524.70	Joback Method

dvisc	0.0001961	Paxs	583.85	Joback Method
dvisc	0.0001168	Paxs	643.00	Joback Method
dvisc	0.0000760	Paxs	702.16	Joback Method
dvisc	0.0000528	Paxs	761.31	Joback Method
dvisc	0.0000387	Paxs	820.46	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370750&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370750&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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