

# Propane, 1,1,3,3-tetramethoxy-

<b>Other names:</b>	Malonaldehyde, bis(dimethyl acetal) Malonaldehyde tetramethyl acetal Tetramethoxypropane 1,1,3,3-Tetramethoxypropane Malondialdehyde bis(dimethylacetal)
<b>Inchi:</b>	InChI=1S/C7H16O4/c1-8-6(9-2)5-7(10-3)11-4/h6-7H,5H2,1-4H3
<b>InchiKey:</b>	XHTYQFMRBQUCPX-UHFFFAOYSA-N
<b>Formula:</b>	C7H16O4
<b>SMILES:</b>	COC(CC(OC)OC)OC
<b>Mol. weight [g/mol]:</b>	164.20
<b>CAS:</b>	102-52-3

## Physical Properties

Property code	Value	Unit	Source
gf	-416.82	kJ/mol	Joback Method
hf	-727.25	kJ/mol	Joback Method
hfus	11.59	kJ/mol	Joback Method
hvap	40.04	kJ/mol	Joback Method
log10ws	-0.32		Crippen Method
logp	0.614		Crippen Method
mcvol	132.970	ml/mol	McGowan Method
pc	2651.56	kPa	Joback Method
rinpol	883.00		NIST Webbook
rinpol	883.00		NIST Webbook
tb	456.20	K	NIST Webbook
tc	620.62	K	Joback Method
tf	227.57	K	Joback Method
vc	0.487	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	289.86	J/molxK	448.36	Joback Method
cpg	301.55	J/molxK	477.07	Joback Method

cpg	313.03	J/mol×K	505.78	Joback Method
cpg	324.27	J/mol×K	534.49	Joback Method
cpg	335.25	J/mol×K	563.20	Joback Method
cpg	345.95	J/mol×K	591.91	Joback Method
cpg	356.36	J/mol×K	620.62	Joback Method
dvisc	0.0036719	Paxs	227.57	Joback Method
dvisc	0.0014372	Paxs	264.37	Joback Method
dvisc	0.0007074	Paxs	301.17	Joback Method
dvisc	0.0004064	Paxs	337.97	Joback Method
dvisc	0.0002603	Paxs	374.76	Joback Method
dvisc	0.0001805	Paxs	411.56	Joback Method
dvisc	0.0001330	Paxs	448.36	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C102523&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C102523&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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