

# 1,3,5-Cycloheptatriene, methyl

<b>Inchi:</b>	InChI=1S/C8H10/c1-8-6-4-2-3-5-7-8/h2-4,6-7H,5H2,1H3
<b>InchiKey:</b>	UQJRWEWFCRBHQV-UHFFFAOYSA-N
<b>Formula:</b>	C8H10
<b>SMILES:</b>	CC1=CCC=CC=C1
<b>Mol. weight [g/mol]:</b>	106.17

## Physical Properties

Property code	Value	Unit	Source
gf	116.79	kJ/mol	Joback Method
hf	21.92	kJ/mol	Joback Method
hfus	8.42	kJ/mol	Joback Method
hvap	35.85	kJ/mol	Joback Method
log10ws	-2.63		Crippen Method
logp	2.449		Crippen Method
mcvol	99.820	ml/mol	McGowan Method
pc	3763.78	kPa	Joback Method
rinpol	890.00		NIST Webbook
rinpol	890.00		NIST Webbook
tb	413.39	K	Joback Method
tc	631.79	K	Joback Method
tf	202.82	K	Joback Method
vc	0.367	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	173.68	J/molxK	413.39	Joback Method
cpg	187.43	J/molxK	449.79	Joback Method
cpg	200.42	J/molxK	486.19	Joback Method
cpg	212.66	J/molxK	522.59	Joback Method
cpg	224.19	J/molxK	558.99	Joback Method
cpg	235.02	J/molxK	595.39	Joback Method
cpg	245.18	J/molxK	631.79	Joback Method
dvisc	0.0055690	Paxs	202.82	Joback Method

dvisc	0.0021470	Paxs	237.91	Joback Method
dvisc	0.0010576	Paxs	273.01	Joback Method
dvisc	0.0006122	Paxs	308.11	Joback Method
dvisc	0.0003962	Paxs	343.20	Joback Method
dvisc	0.0002780	Paxs	378.29	Joback Method
dvisc	0.0002072	Paxs	413.39	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R128124&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R128124&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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