

# 1-Butanol, 4-bromo-

<b>Other names:</b>	4-Bromo-1-butanol 4-bromobutan-1-ol
<b>Inchi:</b>	InChI=1S/C4H9BrO/c5-3-1-2-4-6/h6H,1-4H2
<b>InchiKey:</b>	SIJLYRDVTMMSIP-UHFFFAOYSA-N
<b>Formula:</b>	C4H9BrO
<b>SMILES:</b>	OCCCCBr
<b>Mol. weight [g/mol]:</b>	153.02
<b>CAS:</b>	33036-62-3

## Physical Properties

Property code	Value	Unit	Source
gf	-139.70	kJ/mol	Joback Method
hf	-251.79	kJ/mol	Joback Method
hfus	15.49	kJ/mol	Joback Method
hvap	47.61	kJ/mol	Joback Method
log10ws	-1.19		Crippen Method
logp	1.154		Crippen Method
mvol	90.590	ml/mol	McGowan Method
pc	4815.84	kPa	Joback Method
tb	449.26	K	Joback Method
tc	628.58	K	Joback Method
tf	255.46	K	Joback Method
vc	0.341	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	165.22	J/molxK	449.26	Joback Method
cpg	197.02	J/molxK	598.69	Joback Method
cpg	191.25	J/molxK	568.80	Joback Method
cpg	185.20	J/molxK	538.92	Joback Method
cpg	178.85	J/molxK	509.03	Joback Method
cpg	172.19	J/molxK	479.15	Joback Method
cpg	202.51	J/molxK	628.58	Joback Method

dvisc	0.0003123	Paxs	449.26	Joback Method
dvisc	0.0004967	Paxs	416.96	Joback Method
dvisc	0.0008541	Paxs	384.66	Joback Method
dvisc	0.0016220	Paxs	352.36	Joback Method
dvisc	0.0035062	Paxs	320.06	Joback Method
dvisc	0.0090108	Paxs	287.76	Joback Method
dvisc	0.0294007	Paxs	255.46	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	330.00 ± 1.00	K	0.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33036623&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33036623&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/42-878-1/1-Butanol-4-bromo.pdf>

Generated by Cheméo on 2024-05-03 02:06:27.676645479 +0000 UTC m=+16991236.597222803.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.