

# Cyclohexane, (1-decylundecyl)-

<b>Other names:</b>	Heneicosane, 11-cyclohexyl- 11-Cyclohexylheneicosane
<b>Inchi:</b>	InChI=1S/C27H54/c1-3-5-7-9-11-13-15-18-22-26(27-24-20-17-21-25-27)23-19-16-14-12
<b>InchiKey:</b>	ZSEPDKTYQJRZIS-UHFFFAOYSA-N
<b>Formula:</b>	C27H54
<b>SMILES:</b>	CCCCCCCCCCC(CCCCCCCCCC)C1CCCCC1
<b>Mol. weight [g/mol]:</b>	378.72
<b>CAS:</b>	6703-99-7

## Physical Properties

Property code	Value	Unit	Source
chl	-17651.90 ± 7.10	kJ/mol	NIST Webbook
chl	-17652.30 ± 7.10	kJ/mol	NIST Webbook
gf	198.47	kJ/mol	Joback Method
hf	-551.57	kJ/mol	Joback Method
hfl	-690.70 ± 0.30	kJ/mol	NIST Webbook
hfl	-689.90 ± 7.20	kJ/mol	NIST Webbook
hfus	54.00	kJ/mol	Joback Method
hvap	75.74	kJ/mol	Joback Method
log10ws	-10.54		Crippen Method
logp	10.244		Crippen Method
mcvol	380.430	ml/mol	McGowan Method
pc	772.89	kPa	Joback Method
tb	836.27	K	Joback Method
tc	1025.41	K	Joback Method
tf	266.00 ± 0.70	K	NIST Webbook
tf	266.50 ± 0.70	K	NIST Webbook
tf	265.95 ± 0.40	K	NIST Webbook
tf	265.95 ± 0.50	K	NIST Webbook
tt	294.30 ± 0.20	K	NIST Webbook
vc	1.474	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1412.68	J/mol×K	1025.41	Joback Method
cpg	1283.58	J/mol×K	836.27	Joback Method
cpg	1308.37	J/mol×K	867.79	Joback Method
cpg	1331.77	J/mol×K	899.32	Joback Method
cpg	1353.84	J/mol×K	930.84	Joback Method
cpg	1374.64	J/mol×K	962.36	Joback Method
cpg	1394.23	J/mol×K	993.89	Joback Method
dvisc	0.0000360	Paxs	836.27	Joback Method
dvisc	0.0023926	Paxs	386.43	Joback Method
dvisc	0.0006735	Paxs	461.40	Joback Method
dvisc	0.0002702	Paxs	536.38	Joback Method
dvisc	0.0001356	Paxs	611.35	Joback Method
dvisc	0.0000791	Paxs	686.32	Joback Method
dvisc	0.0000513	Paxs	761.30	Joback Method
hvapt	107.00	kJ/mol	507.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6703997&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6703997&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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