

# Adamantane

<b>Other names:</b>	4,7-methano-1H-indene, octahydro-, (3a.alpha.,4.beta.,7.beta.,7a.alpha.)- 4,7-methanoindan, hexahydro-, exo- JP-10 Tricyclo(3,3,1,1, exo-3,4,8,9-tetrahydrodicyclopentadiene exo-5,6-trimethylenenorbornane exo-octahydro-4,7-methano-1H-indene exo-tetrahydrobicyclopentadiene exo-tetrahydrodicyclopentadiene exo-tricyclo[5.2.1.02,6]decane exo-trimethylenenorbornane jp10 tricyclo[3.3.1.1(3,7)]decane
<b>Inchi:</b>	InChI=1S/C10H16/c1-7-2-9-4-8(1)5-10(3-7)6-9/h7-10H,1-6H2
<b>InchiKey:</b>	ORILYTVJVMKLC-UHFFFAOYSA-N
<b>Formula:</b>	C10H16
<b>SMILES:</b>	C1C2CC3CC1CC(C2)C3
<b>Mol. weight [g/mol]:</b>	136.23
<b>CAS:</b>	281-23-2

## Physical Properties

Property code	Value	Unit	Source
chs	-6033.10 ± 2.80	kJ/mol	NIST Webbook
chs	-6033.40 ± 3.30	kJ/mol	NIST Webbook
chs	-6024.50 ± 0.71	kJ/mol	NIST Webbook
chs	-6029.00 ± 4.00	kJ/mol	NIST Webbook
gf	195.76	kJ/mol	Joback Method
hf	-137.90 ± 0.79	kJ/mol	NIST Webbook
hf	-129.00 ± 4.00	kJ/mol	NIST Webbook
hf	-133.60 ± 2.50	kJ/mol	NIST Webbook
hf	-138.70	kJ/mol	NIST Webbook
hf	-134.40 ± 2.30	kJ/mol	NIST Webbook
hfs	-192.50 ± 0.40	kJ/mol	NIST Webbook
hfs	-188.40 ± 3.30	kJ/mol	NIST Webbook
hfs	-193.80	kJ/mol	NIST Webbook
hfs	-188.70 ± 2.80	kJ/mol	NIST Webbook
hfs	-197.20 ± 0.79	kJ/mol	NIST Webbook

hfus	3.09			kJ/mol	Redetermination of the thermodynamic properties of the solid solid transition of adamantane by adiabatic calorimetry to investigate the suitability as a reference material for low-temperature DSC-calibration
hvap	51.70			kJ/mol	NIST Webbook
hvap	48.20			kJ/mol	NIST Webbook
ie	9.75 ± 0.02			eV	NIST Webbook
ie	9.25			eV	NIST Webbook
ie	9.23			eV	NIST Webbook
ie	9.10 ± 0.05			eV	NIST Webbook
ie	9.30 ± 0.10			eV	NIST Webbook
ie	9.31 ± 0.01			eV	NIST Webbook
ie	9.30 ± 0.01			eV	NIST Webbook
ie	9.22			eV	NIST Webbook
ie	9.55			eV	NIST Webbook
ie	9.25			eV	NIST Webbook
ie	9.75			eV	NIST Webbook
ie	9.75			eV	NIST Webbook
ie	9.28			eV	NIST Webbook
ie	9.25 ± 0.04			eV	NIST Webbook
ie	9.20			eV	NIST Webbook
log10ws	-2.73				Crippen Method
logp	2.833				Crippen Method
mcvol	119.180			ml/mol	McGowan Method
pc	3100.18			kPa	Joback Method
rinpol	1120.00				NIST Webbook
rinpol	1132.00				NIST Webbook
rinpol	1153.00				NIST Webbook
rinpol	1123.00				NIST Webbook
rinpol	1123.00				NIST Webbook
rinpol	1076.00				NIST Webbook
rinpol	1118.00				NIST Webbook
rinpol	1118.00				NIST Webbook
rinpol	1144.00				NIST Webbook
rinpol	1076.00				NIST Webbook
rinpol	1143.00				NIST Webbook
rinpol	1114.00				NIST Webbook
rinpol	1104.00				NIST Webbook
rinpol	1095.00				NIST Webbook
rinpol	1130.00				NIST Webbook
rinpol	1131.00				NIST Webbook
rinpol	1128.00				NIST Webbook

rinpol	1125.00		NIST Webbook
rinpol	1120.00		NIST Webbook
rinpol	1117.00		NIST Webbook
rinpol	1116.00		NIST Webbook
rinpol	1114.00		NIST Webbook
rinpol	1104.00		NIST Webbook
rinpol	1095.00		NIST Webbook
rinpol	1116.00		NIST Webbook
rinpol	1118.00		NIST Webbook
rinpol	1120.00		NIST Webbook
rinpol	1117.00		NIST Webbook
rinpol	1113.00		NIST Webbook
rinpol	1095.20		NIST Webbook
rinpol	1135.30		NIST Webbook
rinpol	1149.00		NIST Webbook
rinpol	1116.00		NIST Webbook
rinpol	1116.00		NIST Webbook
rinpol	1149.00		NIST Webbook
rinpol	1120.00		NIST Webbook
rinpol	1120.00		NIST Webbook
rinpol	1117.00		NIST Webbook
rinpol	1121.00		NIST Webbook
rinpol	1136.90		NIST Webbook
rinpol	1118.00		NIST Webbook
rinpol	1116.00		NIST Webbook
ripol	1400.00		NIST Webbook
ripol	1363.00		NIST Webbook
ripol	1400.00		NIST Webbook
ripol	1363.00		NIST Webbook
ripol	1339.00		NIST Webbook
ripol	1320.00		NIST Webbook
ripol	1363.00		NIST Webbook
ripol	1325.00		NIST Webbook
ripol	1325.00		NIST Webbook
ripol	1400.00		NIST Webbook
ss	195.83	J/molxK	NIST Webbook
ss	195.83	J/molxK	NIST Webbook
tb	448.02	K	Joback Method
tc	661.62	K	Joback Method
tf	542.15 ± 1.00	K	NIST Webbook
tf	552.00 ± 3.00	K	NIST Webbook
vc	0.458	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	272.54	J/molxK	448.02	Joback Method
cpg	293.60	J/molxK	483.62	Joback Method
cpg	313.20	J/molxK	519.22	Joback Method
cpg	331.44	J/molxK	554.82	Joback Method
cpg	348.41	J/molxK	590.42	Joback Method
cpg	364.21	J/molxK	626.02	Joback Method
cpg	378.92	J/molxK	661.62	Joback Method
cps	189.74	J/molxK	298.15	NIST Webbook
cps	189.74	J/molxK	298.15	NIST Webbook
dvisc	0.0006121	Paxs	248.52	Joback Method
dvisc	0.0007158	Paxs	281.77	Joback Method
dvisc	0.0008100	Paxs	315.02	Joback Method
dvisc	0.0008951	Paxs	348.27	Joback Method
dvisc	0.0009721	Paxs	381.52	Joback Method
dvisc	0.0010419	Paxs	414.77	Joback Method
dvisc	0.0011052	Paxs	448.02	Joback Method
hfust	10.90	kJ/mol	541.20	NIST Webbook
hfust	10.90	kJ/mol	541.00	NIST Webbook
hfust	13.80	kJ/mol	543.20	NIST Webbook
hsubt	53.60	kJ/mol	339.00	NIST Webbook
hsubt	58.30	kJ/mol	308.00	NIST Webbook
hsubt	59.70	kJ/mol	323.00	NIST Webbook
hsubt	55.30	kJ/mol	350.50	NIST Webbook
hsubt	54.30	kJ/mol	413.00	NIST Webbook
hsubt	59.70 ± 0.80	kJ/mol	323.00	NIST Webbook
hsubt	59.50	kJ/mol	360.50	NIST Webbook
hsubt	59.30 ± 0.20	kJ/mol	323.00	NIST Webbook
hsubt	58.60 ± 0.60	kJ/mol	333.00	NIST Webbook
rhoI	888.34	kg/m <sup>3</sup>	353.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes

rho1	935.74	kg/m3	293.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
rho1	931.86	kg/m3	298.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
rho1	927.95	kg/m3	303.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
rho1	920.13	kg/m3	313.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
rho1	912.14	kg/m3	323.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
rho1	904.25	kg/m3	333.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
rho1	880.32	kg/m3	363.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes

rhoI	896.32	kg/m <sup>3</sup>	343.15	Density, Viscosity, Surface Tension, and Refractive Index for Binary Mixtures of 1,3-Dimethyladamantane with Four C10 Alkanes
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## Sources

Redetermination of the thermodynamic properties of the solid solid transition of 1,3-dimethyladamantane in binary mixtures with four C10 alkanes	<a href="https://www.doi.org/10.1016/j.tca.2006.02.036">https://www.doi.org/10.1016/j.tca.2006.02.036</a>
Redetermination of the thermodynamic properties of the solid solid transition of 1,3-dimethyladamantane in binary mixtures with four C10 alkanes	<a href="https://www.doi.org/10.1016/j.fluid.2006.07.015">https://www.doi.org/10.1016/j.fluid.2006.07.015</a>
Redetermination of the thermodynamic properties of the solid solid transition of 1,3-dimethyladamantane in binary mixtures with four C10 alkanes	<a href="https://www.doi.org/10.1021/je4008926">https://www.doi.org/10.1021/je4008926</a>
Redetermination of the thermodynamic properties of the solid solid transition of 1,3-dimethyladamantane in binary mixtures with four C10 alkanes	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Redetermination of the thermodynamic properties of the solid solid transition of 1,3-dimethyladamantane in binary mixtures with four C10 alkanes	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C281232&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C281232&amp;Units=SI</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Solubilities of Adamantane and Diamantane in Pressurized Hot Water: Joback Method:	<a href="https://www.doi.org/10.1021/je700709m">https://www.doi.org/10.1021/je700709m</a>
Solubility of Adamantane in Phosphonium-Based Ionic Liquids: McGowan Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
	<a href="https://www.doi.org/10.1021/je9006772">https://www.doi.org/10.1021/je9006772</a>
	<a href="https://link.springer.com/article/10.1007/BF02311772">https://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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