

# Benzoic acid, 4-butyl-, 4-cyanophenyl ester

<b>Other names:</b>	p-Cyanophenyl p-butylbenzoate 4-Cyanophenyl 4-butylbenzoate 4-Butylbenzoic acid, 4-cyanophenyl ester
<b>Inchi:</b>	InChI=1S/C18H17NO2/c1-2-3-4-14-5-9-16(10-6-14)18(20)21-17-11-7-15(13-19)8-12-17/
<b>InchiKey:</b>	XQRFASOUJIKXRE-UHFFFAOYSA-N
<b>Formula:</b>	C18H17NO2
<b>SMILES:</b>	CCCCc1ccc(C(=O)Oc2ccc(C#N)cc2)cc1
<b>Mol. weight [g/mol]:</b>	279.33
<b>CAS:</b>	38690-77-6

## Physical Properties

Property code	Value	Unit	Source
gf	205.50	kJ/mol	Joback Method
hf	-44.65	kJ/mol	Joback Method
hfus	33.97	kJ/mol	Joback Method
hvap	81.17	kJ/mol	Joback Method
log10ws	-5.55		Crippen Method
logp	4.120		Crippen Method
mcvol	225.780	ml/mol	McGowan Method
pc	1901.92	kPa	Joback Method
rinpol	2385.00		NIST Webbook
tb	852.93	K	Joback Method
tc	1089.64	K	Joback Method
tf	507.65	K	Joback Method
vc	0.877	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	646.65	J/molxK	852.93	Joback Method
cpg	659.47	J/molxK	892.38	Joback Method
cpg	671.16	J/molxK	931.83	Joback Method
cpg	681.77	J/molxK	971.28	Joback Method
cpg	691.34	J/molxK	1010.73	Joback Method

cpg	699.92	J/mol×K	1050.18	Joback Method
cpg	707.57	J/mol×K	1089.64	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C38690776&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C38690776&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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