

1-Octen-3-yne

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| Other names: | 1-Octene-3-yne n-C ₄ H ₉ C«equiv»CCH=CH ₂ n-C ₄ H ₉ CÂ«equivÂ»CCH=CH ₂ |
| Inchi: | InChI=1S/C ₈ H ₁₂ /c1-3-5-7-8-6-4-2/h3H,1,4,6,8H ₂ ,2H ₃ |
| InchiKey: | UEKPKVCWUZMBML-UHFFFAOYSA-N |
| Formula: | C ₈ H ₁₂ |
| SMILES: | C=CC#CCCCC |
| Mol. weight [g/mol]: | 108.18 |
| CAS: | 17679-92-4 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------------|----------------------|----------------|
| gf | 307.12 | kJ/mol | Joback Method |
| hf | 189.28 | kJ/mol | Joback Method |
| hfus | 18.32 | kJ/mol | Joback Method |
| hvap | 34.88 | kJ/mol | Joback Method |
| ie | 8.83 ± 0.01 | eV | NIST Webbook |
| log10ws | -2.82 | | Crippen Method |
| logp | 2.366 | | Crippen Method |
| mcvol | 110.680 | ml/mol | McGowan Method |
| pc | 3142.03 | kPa | Joback Method |
| tb | 407.00 ± 3.00 | K | NIST Webbook |
| tc | 579.51 | K | Joback Method |
| tf | 284.26 | K | Joback Method |
| vc | 0.426 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 193.89 | J/mol×K | 388.12 | Joback Method |
| cpg | 205.28 | J/mol×K | 420.02 | Joback Method |
| cpg | 216.17 | J/mol×K | 451.92 | Joback Method |
| cpg | 226.59 | J/mol×K | 483.82 | Joback Method |
| cpg | 236.55 | J/mol×K | 515.72 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 246.07 | J/mol×K | 547.61 | Joback Method |
| cpg | 255.16 | J/mol×K | 579.51 | Joback Method |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.53674e+01 |
| Coeff. B | -3.79081e+03 |
| Coeff. C | -5.43380e+01 |
| Temperature range (K), min. | 305.72 |
| Temperature range (K), max. | 431.31 |

Sources

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|---|---|
| The Yaws Handbook of Vapor Pressure: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C17679924&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |

| | |
|--------------|----------------------------------|
| pvap: | Vapor pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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