

# 7-Oxabicyclo[2.2.1]heptane

<b>Other names:</b>	7-Oxanorbornane Cyclohexane, 1,4-epoxy- Cyclohexane, 3,6-endoxo- 1,4-Epoxycyclohexane 1,4-Oxycyclohexane 3,6-Endooxycyclohexane 3,6-Endoxocyclohexane Bicyclo[2.2.1]heptane, 7-oxa-
<b>Inchi:</b>	InChI=1S/C6H10O/c1-2-6-4-3-5(1)7-6/h5-6H,1-4H2
<b>InchiKey:</b>	YPWFNLSXQIGJCK-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O
<b>SMILES:</b>	C1CC2CCC1O2
<b>Mol. weight [g/mol]:</b>	98.14
<b>CAS:</b>	279-49-2

## Physical Properties

Property code	Value	Unit	Source
affp	844.20	kJ/mol	NIST Webbook
basg	816.80	kJ/mol	NIST Webbook
chl	-3566.30 ± 1.50	kJ/mol	NIST Webbook
chl	-3566.30 ± 1.50	kJ/mol	NIST Webbook
gf	22.92	kJ/mol	Joback Method
hf	-159.73	kJ/mol	Joback Method
hfl	-223.90 ± 2.00	kJ/mol	NIST Webbook
hfus	13.44	kJ/mol	Joback Method
hvap	33.46	kJ/mol	Joback Method
ie	9.57 ± 0.02	eV	NIST Webbook
log10ws	-1.43		Crippen Method
logp	1.328		Crippen Method
mcvol	79.550	ml/mol	McGowan Method
pc	4339.67	kPa	Joback Method
tb	381.38	K	Joback Method
tc	586.65	K	Joback Method
tf	216.31	K	Joback Method
vc	0.298	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	151.20	J/molxK	381.38	Joback Method
cpg	165.70	J/molxK	415.59	Joback Method
cpg	179.24	J/molxK	449.80	Joback Method
cpg	191.89	J/molxK	484.02	Joback Method
cpg	203.68	J/molxK	518.23	Joback Method
cpg	214.68	J/molxK	552.44	Joback Method
cpg	224.93	J/molxK	586.65	Joback Method
dvisc	0.0009627	Paxs	216.31	Joback Method
dvisc	0.0008301	Paxs	243.82	Joback Method
dvisc	0.0007377	Paxs	271.33	Joback Method
dvisc	0.0006699	Paxs	298.85	Joback Method
dvisc	0.0006183	Paxs	326.36	Joback Method
dvisc	0.0005779	Paxs	353.87	Joback Method
dvisc	0.0005454	Paxs	381.38	Joback Method
hfust	0.71	kJ/mol	244.00	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	392.20	K	95.10	NIST Webbook

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C279492&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C279492&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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