

# Benzene, 1,3-dimethoxy-2,4-dimethyl

<b>Inchi:</b>	InChI=1S/C10H14O2/c1-7-5-6-9(11-3)8(2)10(7)12-4/h5-6H,1-4H3
<b>InchiKey:</b>	LPAMMSFOMIQXDN-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O2
<b>SMILES:</b>	COc1ccc(C)c(OC)c1C
<b>Mol. weight [g/mol]:</b>	166.22

## Physical Properties

Property code	Value	Unit	Source
gf	-93.16	kJ/mol	Joback Method
hf	-312.05	kJ/mol	Joback Method
hfus	16.91	kJ/mol	Joback Method
hvap	46.94	kJ/mol	Joback Method
log10ws	-2.66		Crippen Method
logp	2.321		Crippen Method
mvol	139.740	ml/mol	McGowan Method
pc	2695.80	kPa	Joback Method
rinpol	1283.00		NIST Webbook
rinpol	1283.00		NIST Webbook
tb	514.66	K	Joback Method
tc	719.69	K	Joback Method
tf	310.90	K	Joback Method
vc	0.523	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	306.26	J/molxK	514.66	Joback Method
cpg	319.60	J/molxK	548.83	Joback Method
cpg	332.44	J/molxK	583.00	Joback Method
cpg	344.78	J/molxK	617.18	Joback Method
cpg	356.59	J/molxK	651.35	Joback Method
cpg	367.88	J/molxK	685.52	Joback Method
cpg	378.62	J/molxK	719.69	Joback Method
dvisc	0.0008526	Paxs	310.90	Joback Method

dvisc	0.0005514	Paxs	344.86	Joback Method
dvisc	0.0003856	Paxs	378.82	Joback Method
dvisc	0.0002860	Paxs	412.78	Joback Method
dvisc	0.0002220	Paxs	446.74	Joback Method
dvisc	0.0001786	Paxs	480.70	Joback Method
dvisc	0.0001478	Paxs	514.66	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R142759&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R142759&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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<https://www.chemeo.com/cid/40-646-0/Benzene-1-3-dimethoxy-2-4-dimethyl.pdf>

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