

# Propanenitrile, 3-chloro-

<b>Other names:</b>	1-Chloro-2-cyanoethane 3-Chloropropanenitrile 3-Chloropropanonitrile 3-Chloropropionitrile 3-Chlorpropannitril 3-chloroprop-2-ynenitrile 3-chloropropiononitrile CH <sub>2</sub> ClCH <sub>2</sub> CN Cl(CH <sub>2</sub> ) <sub>2</sub> CN NSC 2588 Propionitrile, 3-chloro- Rcra waste number P027 USAF A-8798 «beta»-Chloropropionitrile Â«betaÂ»-Chloropropionitrile
<b>Inchi:</b>	InChI=1S/C3H4ClN/c4-2-1-3-5/h1-2H2
<b>InchiKey:</b>	GNHMRTZZNHZDDM-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>3</sub> H <sub>4</sub> ClN
<b>SMILES:</b>	N#CCCCl
<b>Mol. weight [g/mol]:</b>	89.52
<b>CAS:</b>	542-76-7

## Physical Properties

Property code	Value	Unit	Source
affp	773.10	kJ/mol	NIST Webbook
basg	742.40	kJ/mol	NIST Webbook
gf	95.63	kJ/mol	Joback Method
hf	43.89	kJ/mol	Joback Method
hfus	9.23	kJ/mol	Joback Method
hvap	37.14	kJ/mol	Joback Method
log10ws	-0.29		Estimated Solubility Method
log10ws	-0.29		Aqueous Solubility Prediction Method
logp	1.139		Crippen Method
mcvol	66.750	ml/mol	McGowan Method
pc	4067.32	kPa	Joback Method

tb	448.70	K	NIST Webbook
tc	609.29	K	Joback Method
tf	222.48	K	Aqueous Solubility Prediction Method
tf	221.70 ± 0.10	K	NIST Webbook
tf	221.63 ± 0.20	K	NIST Webbook
vc	0.279	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	104.40	J/mol×K	407.55	Joback Method
cpg	108.97	J/mol×K	441.17	Joback Method
cpg	113.32	J/mol×K	474.80	Joback Method
cpg	117.47	J/mol×K	508.42	Joback Method
cpg	121.43	J/mol×K	542.05	Joback Method
cpg	125.19	J/mol×K	575.67	Joback Method
cpg	128.77	J/mol×K	609.29	Joback Method

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:** [http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C542767&Units=SI>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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