

# Pantolactone

<b>Other names:</b>	(D)-Pantolactone (R)-(-)-Pantolactone (R)-3-hydroxy-4,4-dimethyldihydrofuran-2(3H)-one (R)-pantolactone 2(3H)-Furanone, dihydro-3-hydroxy-4,4-dimethyl-, (3R)- 2(3H)-Furanone, dihydro-3-hydroxy-4,4-dimethyl-, (R)- 2(3H)-Furanone, dihydro-3-hydroxy-4,4-dimethyl-, D-(-)- 2-hydroxy-3,3-dimethyl-«gamma»-butyrolactone D(-)-2-Hydroxy-3,3-dimethyl-«gamma»-butyrolactone D-(-)-Pantoic acid lactone D-(-)-Pantolactone D-(-)-Pantolyl lactone D-(-)-Pantoyl lactone D-(-)-«alpha»Hydroxy-«beta», «beta»-dimethyl-«gamma»-butyrolactone D-pantolactone Dihydro-3-hydroxy-4,4-dimethyl-2(3H)-furanone, R-(-)-pantolactone Dihydro-3-hydroxy-4,4-dimethyl-2(3H)-furanone, pantolactone Pantolyl lactone Pantothenic lactone Pantoyl lactone «alpha»-hydroxy-«beta», «beta»-dimethyl-«gamma»-butyrolactone
<b>Inchi:</b>	InChI=1S/C6H10O3/c1-6(2)3-9-5(8)4(6)7/h4,7H,3H2,1-2H3/t4-/m1/s1
<b>InchiKey:</b>	SERHXTVXHNVDKA-SCSAIBSYSA-N
<b>Formula:</b>	C6H10O3
<b>SMILES:</b>	CC1(C)COC(=O)C1O
<b>Mol. weight [g/mol]:</b>	130.14
<b>CAS:</b>	599-04-2

## Physical Properties

Property code	Value	Unit	Source
gf	-322.54	kJ/mol	Joback Method
hf	-533.72	kJ/mol	Joback Method
hfus	11.58	kJ/mol	Joback Method
hvap	53.18	kJ/mol	Joback Method
log10ws	-0.22		Crippen Method
logp	-0.070		Crippen Method

mvol	97.850	ml/mol	McGowan Method
pc	4540.80	kPa	Joback Method
ripol	2034.00		NIST Webbook
ripol	1998.00		NIST Webbook
ripol	1998.00		NIST Webbook
ripol	2033.00		NIST Webbook
ripol	2051.00		NIST Webbook
ripol	2077.00		NIST Webbook
ripol	2033.00		NIST Webbook
ripol	2077.00		NIST Webbook
ripol	2008.00		NIST Webbook
ripol	2006.00		NIST Webbook
ripol	2028.00		NIST Webbook
ripol	2051.00		NIST Webbook
ripol	2006.00		NIST Webbook
ripol	2029.00		NIST Webbook
ripol	2028.00		NIST Webbook
tb	534.48	K	Joback Method
tc	745.23	K	Joback Method
tf	343.55	K	Joback Method
vc	0.356	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.67	J/mol×K	534.48	Joback Method
cpg	253.86	J/mol×K	569.60	Joback Method
cpg	264.47	J/mol×K	604.73	Joback Method
cpg	274.55	J/mol×K	639.85	Joback Method
cpg	284.18	J/mol×K	674.98	Joback Method
cpg	293.41	J/mol×K	710.10	Joback Method
cpg	302.32	J/mol×K	745.23	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	394.20	K	2.00	NIST Webbook

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Experimental and Modeling Studies on the Solubility of d-Pantolactone in Four Polar Solvents and Ethanol Water Mixtures:</b>	<a href="https://www.doi.org/10.1021/je500996n">https://www.doi.org/10.1021/je500996n</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C599042&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C599042&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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