

# Fumaric acid, pent-4-en-2-yl propyl ester

<b>Inchi:</b>	InChI=1S/C12H18O4/c1-4-6-10(3)16-12(14)8-7-11(13)15-9-5-2/h4,7-8,10H,1,5-6,9H2,2-3
<b>InchiKey:</b>	FFCNSQCWMWSGEA-BQYQJAHWSA-N
<b>Formula:</b>	C12H18O4
<b>SMILES:</b>	C=CCC(C)OC(=O)C=CC(=O)OCCC
<b>Mol. weight [g/mol]:</b>	226.27

## Physical Properties

Property code	Value	Unit	Source
gf	-252.06	kJ/mol	Joback Method
hf	-543.24	kJ/mol	Joback Method
hfus	27.81	kJ/mol	Joback Method
hvap	59.52	kJ/mol	Joback Method
log10ws	-2.39		Crippen Method
logp	2.004		Crippen Method
mcvol	186.220	ml/mol	McGowan Method
pc	2111.94	kPa	Joback Method
rinsol	1504.00		NIST Webbook
tb	626.94	K	Joback Method
tc	816.65	K	Joback Method
tf	347.48	K	Joback Method
vc	0.711	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	473.65	J/molxK	626.94	Joback Method
cpg	535.74	J/molxK	785.03	Joback Method
cpg	524.64	J/molxK	753.41	Joback Method
cpg	512.90	J/molxK	721.79	Joback Method
cpg	500.50	J/molxK	690.18	Joback Method
cpg	487.42	J/molxK	658.56	Joback Method
cpg	546.19	J/molxK	816.65	Joback Method
dvisc	0.0001266	Paxs	626.94	Joback Method
dvisc	0.0001662	Paxs	580.36	Joback Method

dvisc	0.0002288	Paxs	533.79	Joback Method
dvisc	0.0003348	Paxs	487.21	Joback Method
dvisc	0.0005310	Paxs	440.63	Joback Method
dvisc	0.0009391	Paxs	394.06	Joback Method
dvisc	0.0019354	Paxs	347.48	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348921&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348921&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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