

# 2-Butanone, ethylhydrazone

<b>Other names:</b>	Methylethylketone ethylhydrazone
<b>Inchi:</b>	InChI=1S/C6H14N2/c1-4-6(3)8-7-5-2/h7H,4-5H2,1-3H3
<b>InchiKey:</b>	PCKXYWXISYJOTB-UHFFFAOYSA-N
<b>Formula:</b>	C6H14N2
<b>SMILES:</b>	CCNN=C(C)CC
<b>Mol. weight [g/mol]:</b>	114.19
<b>CAS:</b>	16713-35-2

## Physical Properties

Property code	Value	Unit	Source
hf	-41.27	kJ/mol	Joback Method
hvap	38.78	kJ/mol	Joback Method
log10ws	-1.68		Crippen Method
logp	1.382		Crippen Method
mcvol	111.060	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
rinpol	881.00		NIST Webbook
rinpol	881.00		NIST Webbook
tb	463.41	K	Joback Method
tc	658.15	K	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16713352&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16713352&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinqol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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