

# Benzene, 1-nitro-4-(phenylmethyl)-

<b>Other names:</b>	Methane, (p-nitrophenyl)phenyl- (p-Nitrophenyl)phenylmethane p-Nitrodiphenylmethane 4-Nitrodiphenylmethane
<b>Inchi:</b>	InChI=1S/C13H11NO2/c15-14(16)13-8-6-12(7-9-13)10-11-4-2-1-3-5-11/h1-9H,10H2
<b>InchiKey:</b>	IDSGFSCSMXRJON-UHFFFAOYSA-N
<b>Formula:</b>	C13H11NO2
<b>SMILES:</b>	O=[N+]([O-])c1ccc(Cc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	213.23
<b>CAS:</b>	1817-77-2

## Physical Properties

Property code	Value	Unit	Source
gf	309.32	kJ/mol	Joback Method
hf	139.18	kJ/mol	Joback Method
hfus	28.48	kJ/mol	Joback Method
hvap	66.34	kJ/mol	Joback Method
ie	9.35 ± 0.05	eV	NIST Webbook
log10ws	-4.22		Crippen Method
logp	3.186		Crippen Method
mcvol	163.930	ml/mol	McGowan Method
pc	3062.55	kPa	Joback Method
rinpol	1933.00		NIST Webbook
rinpol	1893.30		NIST Webbook
rinpol	1933.00		NIST Webbook
rinpol	1893.30		NIST Webbook
tb	707.02	K	Joback Method
tc	975.37	K	Joback Method
tf	445.24	K	Joback Method
vc	0.629	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	423.97	J/mol×K	707.02	Joback Method
cpg	438.23	J/mol×K	751.74	Joback Method
cpg	451.16	J/mol×K	796.47	Joback Method
cpg	462.88	J/mol×K	841.19	Joback Method
cpg	473.48	J/mol×K	885.92	Joback Method
cpg	483.06	J/mol×K	930.64	Joback Method
cpg	491.73	J/mol×K	975.37	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1817772&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1817772&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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