

# 2-Bromo-1-chloro-2-methylpropane

<b>Other names:</b>	Propane, 2-bromo-1-chloro-2-methyl-
<b>Inchi:</b>	InChI=1S/C4H8BrCl/c1-4(2,5)3-6/h3H2,1-2H3
<b>InchiKey:</b>	YEZMNTBSGDUYBT-UHFFFAOYSA-N
<b>Formula:</b>	C4H8BrCl
<b>SMILES:</b>	CC(C)(Br)CCl
<b>Mol. weight [g/mol]:</b>	171.46
<b>CAS:</b>	2074-80-8

## Physical Properties

Property code	Value	Unit	Source
gf	-11.97	kJ/mol	Joback Method
hf	-124.05	kJ/mol	Joback Method
hfus	8.18	kJ/mol	Joback Method
hvap	34.02	kJ/mol	Joback Method
log10ws	-2.19		Crippen Method
logp	2.399		Crippen Method
mcvol	96.960	ml/mol	McGowan Method
pc	4156.97	kPa	Joback Method
tb	391.28	K	Joback Method
tc	599.33	K	Joback Method
tf	226.98	K	Joback Method
vc	0.359	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	151.92	J/molxK	391.28	Joback Method
cpg	160.82	J/molxK	425.96	Joback Method
cpg	169.09	J/molxK	460.63	Joback Method
cpg	176.76	J/molxK	495.31	Joback Method
cpg	183.87	J/molxK	529.98	Joback Method
cpg	190.46	J/molxK	564.66	Joback Method
cpg	196.57	J/molxK	599.33	Joback Method
dvisc	0.0058378	Paxs	226.98	Joback Method

dvisc	0.0030285	Paxs	254.36	Joback Method
dvisc	0.0017849	Paxs	281.75	Joback Method
dvisc	0.0011552	Paxs	309.13	Joback Method
dvisc	0.0008026	Paxs	336.51	Joback Method
dvisc	0.0005890	Paxs	363.90	Joback Method
dvisc	0.0004514	Paxs	391.28	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2074808&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2074808&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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