

# isobutyl-n-butyl-amine

<b>Other names:</b>	Butylisobutyl amine N-isobutylbutylamine 1-Butanamine, N-(2-methylpropyl)-
<b>Inchi:</b>	InChI=1S/C8H19N/c1-4-5-6-9-7-8(2)3/h8-9H,4-7H2,1-3H3
<b>InchiKey:</b>	CCRUHKGZAQXBKC-UHFFFAOYSA-N
<b>Formula:</b>	C8H19N
<b>SMILES:</b>	CCCCNCC(C)C
<b>Mol. weight [g/mol]:</b>	129.24
<b>CAS:</b>	20810-06-4

## Physical Properties

Property code	Value	Unit	Source
chl	-5647.70 ± 5.20	kJ/mol	NIST Webbook
gf	103.43	kJ/mol	Joback Method
hf	-175.00 ± 6.30	kJ/mol	NIST Webbook
hfl	-216.00 ± 5.00	kJ/mol	NIST Webbook
hfus	18.05	kJ/mol	Joback Method
hvap	41.00	kJ/mol	NIST Webbook
log10ws	-2.12		Crippen Method
logp	2.032		Crippen Method
mcvol	133.560	ml/mol	McGowan Method
pc	2589.85	kPa	Joback Method
rinpol	898.00		NIST Webbook
ripol	1014.00		NIST Webbook
ripol	1014.00		NIST Webbook
tb	432.17	K	Joback Method
tc	604.23	K	Joback Method
tf	217.58	K	Joback Method
vc	0.512	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	332.20	J/mol×K	546.88	Joback Method

cpg	344.10	J/mol×K	575.56	Joback Method
cpg	279.40	J/mol×K	432.17	Joback Method
cpg	293.41	J/mol×K	460.85	Joback Method
cpg	306.87	J/mol×K	489.52	Joback Method
cpg	319.80	J/mol×K	518.20	Joback Method
cpg	355.50	J/mol×K	604.23	Joback Method
hvapt	41.20	kJ/mol	368.00	NIST Webbook
hvapt	41.00 ± 1.00	kJ/mol	313.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20810064&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20810064&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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