

# Tricosanoic acid

<b>Other names:</b>	n-Tricosanoic acid
<b>Inchi:</b>	InChI=1S/C23H46O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23(24)
<b>InchiKey:</b>	XEZVDURJDFGERA-UHFFFAOYSA-N
<b>Formula:</b>	C23H46O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	354.61
<b>CAS:</b>	2433-96-7

## Physical Properties

Property code	Value	Unit	Source
gf	-122.96	kJ/mol	Joback Method
hf	-782.86	kJ/mol	Joback Method
hfus	61.01	kJ/mol	Joback Method
hvap	90.22	kJ/mol	Joback Method
log10ws	-8.55		Crippen Method
logp	8.283		Crippen Method
mcvol	342.370	ml/mol	McGowan Method
pc	934.63	kPa	Joback Method
rinpol	2668.10		NIST Webbook
tb	871.69	K	Joback Method
tc	1069.35	K	Joback Method
tf	324.40 ± 3.00	K	NIST Webbook
tf	339.00 ± 4.00	K	NIST Webbook
tf	351.90 ± 2.00	K	NIST Webbook
vc	1.349	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1246.66	J/mol×K	1069.35	Joback Method
cpg	1231.46	J/mol×K	1036.41	Joback Method
cpg	1215.36	J/mol×K	1003.46	Joback Method
cpg	1198.29	J/mol×K	970.52	Joback Method
cpg	1180.19	J/mol×K	937.58	Joback Method

cpg	1161.00	J/mol×K	904.63	Joback Method
cpg	1140.68	J/mol×K	871.69	Joback Method
dvisc	0.0009361	Paxs	459.72	Joback Method
dvisc	0.0000097	Paxs	871.69	Joback Method
dvisc	0.0000150	Paxs	803.03	Joback Method
dvisc	0.0000252	Paxs	734.37	Joback Method
dvisc	0.0000471	Paxs	665.70	Joback Method
dvisc	0.0001014	Paxs	597.04	Joback Method
dvisc	0.0002667	Paxs	528.38	Joback Method
hfust	75.00	kJ/mol	352.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.04425e+01
Coeff. B	-8.65242e+03
Coeff. C	-1.48024e+02
Temperature range (K), min.	577.32
Temperature range (K), max.	719.86

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2433967&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2433967&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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