

# 2-Butenoic acid, methyl ester, (Z)-

<b>Other names:</b>	Crotonic acid, methyl ester, (Z)- Methyl (Z)-2-butenoate Methyl cis-2-butenoate
<b>Inchi:</b>	InChI=1S/C5H8O2/c1-3-4-5(6)7-2/h3-4H,1-2H3/b4-3-
<b>InchiKey:</b>	MCVVUJXPXSBQTRZ-ARJAWSKDSA-N
<b>Formula:</b>	C5H8O2
<b>SMILES:</b>	CC=CC(=O)OC
<b>Mol. weight [g/mol]:</b>	100.12
<b>CAS:</b>	4358-59-2

## Physical Properties

Property code	Value	Unit	Source
gf	-162.48	kJ/mol	Joback Method
hf	-274.11	kJ/mol	Joback Method
hfus	11.69	kJ/mol	Joback Method
hvap	35.84	kJ/mol	Joback Method
log10ws	-0.63		Crippen Method
logp	0.735		Crippen Method
mcvol	84.450	ml/mol	McGowan Method
pc	3906.25	kPa	Joback Method
rinpol	726.00		NIST Webbook
rinpol	735.00		NIST Webbook
rinpol	726.00		NIST Webbook
ripol	1032.00		NIST Webbook
ripol	1032.00		NIST Webbook
tb	394.25	K	Joback Method
tc	582.22	K	Joback Method
tf	213.19	K	Joback Method
vc	0.320	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	148.24	J/molxK	394.25	Joback Method

cpg	156.32	J/molxK	425.58	Joback Method
cpg	164.08	J/molxK	456.91	Joback Method
cpg	171.53	J/molxK	488.23	Joback Method
cpg	178.67	J/molxK	519.56	Joback Method
cpg	185.52	J/molxK	550.89	Joback Method
cpg	192.07	J/molxK	582.22	Joback Method
dvisc	0.0025463	Paxs	213.19	Joback Method
dvisc	0.0013230	Paxs	243.37	Joback Method
dvisc	0.0007942	Paxs	273.54	Joback Method
dvisc	0.0005276	Paxs	303.72	Joback Method
dvisc	0.0003774	Paxs	333.90	Joback Method
dvisc	0.0002854	Paxs	364.07	Joback Method
dvisc	0.0002253	Paxs	394.25	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.83123e+01
Coeff. B	-4.39316e+03
Coeff. C	-4.98900e+01
Temperature range (K), min.	293.62
Temperature range (K), max.	387.80

## Sources

**The Yaws Handbook of Vapor**

**Pressure:**  
**Crippen Method:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C4358592&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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