

Acetyl chloride, chloro-

Other names:	ALPHA-CHLOROACETYL CHLORIDE Acetyl chloride, 2-chloro- CH ₂ ClCOCl CHLOROACETIC ACID CHLORIDE Chloracetyl chloride Chlorid kyseliny chloroctove Chloroacetic chloride Chloroacetyl chloride Chloroethanoyl chloride Chlorure de chloracetyle Monochloroacetyl chloride UN 1752 «alpha»-Chloroacetyl chloride Â«alphaÂ»-Chloroacetyl chloride
Inchi:	InChI=1S/C2H2Cl2O/c3-1-2(4)5/h1H2
InchiKey:	VG CXGMAHQTYDJK-UHFFFAOYSA-N
Formula:	C ₂ H ₂ Cl ₂ O
SMILES:	O=C(Cl)CCl
Mol. weight [g/mol]:	112.94
CAS:	79-04-9

Physical Properties

Property code	Value	Unit	Source
gf	-186.82	kJ/mol	Joback Method
hf	-246.00 ± 8.80	kJ/mol	NIST Webbook
hfl	-285.00 ± 8.40	kJ/mol	NIST Webbook
hfus	10.93	kJ/mol	Joback Method
hvap	35.56	kJ/mol	Joback Method
ie	10.30	eV	NIST Webbook
log10ws	-0.74		Crippen Method
logp	0.991		Crippen Method
mcpvol	65.090	ml/mol	McGowan Method
pc	4952.36	kPa	Joback Method
rinpol	622.00		NIST Webbook
rinpol	622.00		NIST Webbook
rinpol	622.00		NIST Webbook
tb	373.89	K	Joback Method

tc	572.48	K	Joback Method
tf	377.00	K	NIST Webbook
tf	251.38 ± 0.07	K	NIST Webbook
vc	0.252	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	87.44	J/mol×K	373.89	Joback Method
cpg	90.94	J/mol×K	406.99	Joback Method
cpg	94.26	J/mol×K	440.09	Joback Method
cpg	97.40	J/mol×K	473.19	Joback Method
cpg	100.38	J/mol×K	506.28	Joback Method
cpg	103.20	J/mol×K	539.38	Joback Method
cpg	105.86	J/mol×K	572.48	Joback Method
dvisc	0.0019197	Paxs	247.37	Joback Method
dvisc	0.0031682	Paxs	222.07	Joback Method
dvisc	0.0012766	Paxs	272.68	Joback Method
dvisc	0.0009098	Paxs	297.98	Joback Method
dvisc	0.0006838	Paxs	323.28	Joback Method
dvisc	0.0005356	Paxs	348.59	Joback Method
dvisc	0.0004337	Paxs	373.89	Joback Method
hvapt	45.00	kJ/mol	316.00	NIST Webbook
hvapt	40.80	kJ/mol	340.50	NIST Webbook
hvapt	44.10	kJ/mol	331.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47993e+01
Coeff. B	-3.27151e+03
Coeff. C	-5.76630e+01
Temperature range (K), min.	283.10
Temperature range (K), max.	402.68

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C79049&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.cheric.org/files/research/kdb/mol/mol1772.mol

Legend

cp_g:	Ideal gas heat capacity
d_{visc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fl}:	Liquid phase enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log_{10ws}:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
pv_{ap}:	Vapor pressure
rin_{pol}:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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