

# Formamide, N,N-diphenyl-

<b>Other names:</b>	Formanilide, N-phenyl- N,N-Diphenylformamide Diphenylformamide N-Formyldiphenylamine
<b>Inchi:</b>	InChI=1S/C13H11NO/c15-11-14(12-7-3-1-4-8-12)13-9-5-2-6-10-13/h1-11H
<b>InchiKey:</b>	DCNUQRBLZWSGAV-UHFFFAOYSA-N
<b>Formula:</b>	C13H11NO
<b>SMILES:</b>	O=CN(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	197.23
<b>CAS:</b>	607-00-1

## Physical Properties

Property code	Value	Unit	Source
gf	294.66	kJ/mol	Joback Method
hf	143.36	kJ/mol	Joback Method
hfus	22.82	kJ/mol	Joback Method
hvap	57.85	kJ/mol	Joback Method
log10ws	-2.95		Crippen Method
logp	2.981		Crippen Method
mcvol	158.060	ml/mol	McGowan Method
pc	3333.53	kPa	Joback Method
tb	610.70	K	NIST Webbook
tc	853.47	K	Joback Method
tf	363.58	K	Joback Method
vc	0.583	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	379.27	J/molxK	611.30	Joback Method
cpg	394.58	J/molxK	651.66	Joback Method
cpg	408.57	J/molxK	692.02	Joback Method
cpg	421.34	J/molxK	732.38	Joback Method
cpg	432.98	J/molxK	772.75	Joback Method

cpg	443.59	J/mol×K	813.11	Joback Method
cpg	453.24	J/mol×K	853.47	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	610.20	K	102.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C607001&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C607001&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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