

samarium

Inchi:	InChI=1S/Sm
InchiKey:	KZUNJOHGZRPMI-UHFFFAOYSA-N
Formula:	Sm
SMILES:	[Sm]
Mol. weight [g/mol]:	150.36
CAS:	7440-19-9

Physical Properties

Property code	Value	Unit	Source
ie	5.64 ± 0.00	eV	NIST Webbook
ie	5.64	eV	NIST Webbook
ie	5.64 ± 0.00	eV	NIST Webbook
ie	5.64 ± 0.00	eV	NIST Webbook
ie	5.64 ± 0.00	eV	NIST Webbook
ie	5.64 ± 0.00	eV	NIST Webbook
ie	5.50	eV	NIST Webbook
ie	5.60 ± 0.10	eV	NIST Webbook
ie	5.61 ± 0.05	eV	NIST Webbook
ie	5.56 ± 0.10	eV	NIST Webbook
ie	5.63 ± 0.02	eV	NIST Webbook
ie	6.15 ± 0.05	eV	NIST Webbook
ie	5.70 ± 0.02	eV	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{\text{vap}}) = A + B/(T + C)$
Coeff. A	1.34681e+01
Coeff. B	-1.66773e+04
Coeff. C	-1.82670e+02
Temperature range (K), min.	973.15
Temperature range (K), max.	2273.15

Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7440199&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Synthesis, structure, and thermodynamics of a lanthanide thermodynamic stability of BN₂La₂O₃ phases.
Investigation in the variation of Gibbs energy of formation of RE₆UO₁₂ (RE = La, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Tm, Yb, Lu) along the 4f series:

<https://www.doi.org/10.1016/j.jct.2012.06.029>

<https://www.doi.org/10.1016/j.jct.2013.05.044>

<https://www.doi.org/10.1016/j.jct.2019.06.030>

Legend

ie: Ionization energy

p_{vap}: Vapor pressure

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