

# Phenanthrene, 9,10-diphenyl-

<b>Other names:</b>	9,10-Diphenylphenanthrene
<b>Inchi:</b>	InChI=1S/C26H18/c1-3-11-19(12-4-1)25-23-17-9-7-15-21(23)22-16-8-10-18-24(22)26(25)
<b>InchiKey:</b>	SECWSIFCFMVOAN-UHFFFAOYSA-N
<b>Formula:</b>	C26H18
<b>SMILES:</b>	<chem>c1ccc(-c2c(-c3ccccc3)c3ccccc3c3ccccc23)cc1</chem>
<b>Mol. weight [g/mol]:</b>	330.42
<b>CAS:</b>	602-15-3

## Physical Properties

Property code	Value	Unit	Source
gf	689.68	kJ/mol	Joback Method
hf	477.35	kJ/mol	Joback Method
hfus	38.09	kJ/mol	Joback Method
hvap	85.56	kJ/mol	Joback Method
log10ws	-10.29		Crippen Method
logp	7.327		Crippen Method
mcvol	267.000	ml/mol	McGowan Method
pc	1930.44	kPa	Joback Method
tb	927.22	K	Joback Method
tc	1207.03	K	Joback Method
tf	565.00	K	Joback Method
vc	1.012	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	789.72	J/mol×K	927.22	Joback Method
cpg	864.05	J/mol×K	1160.39	Joback Method
cpg	849.86	J/mol×K	1113.76	Joback Method
cpg	835.60	J/mol×K	1067.12	Joback Method
cpg	821.02	J/mol×K	1020.49	Joback Method
cpg	805.81	J/mol×K	973.85	Joback Method
cpg	878.48	J/mol×K	1207.03	Joback Method
dvisc	0.0002105	Paxs	927.22	Joback Method

dvisc	0.0002466	Paxs	866.85	Joback Method
dvisc	0.0002959	Paxs	806.48	Joback Method
dvisc	0.0003657	Paxs	746.11	Joback Method
dvisc	0.0004691	Paxs	685.74	Joback Method
dvisc	0.0006313	Paxs	625.37	Joback Method
dvisc	0.0009054	Paxs	565.00	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C602153&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C602153&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/33-137-3/Phenanthrene-9-10-diphenyl.pdf>

Generated by Cheméo on 2024-04-27 22:27:52.303234448 +0000 UTC m=+16546121.223811763.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.