

# 2-Propenoic acid, pentyl ester

<b>Other names:</b>	2-Propenoic acid,n-pentyl ester Pentyl acrylate acrylic acid, pentyl ester amyl acrylate n-Amyl acrylate n-pentyl acrylate pentyl 2-propenoate
<b>Inchi:</b>	InChI=1S/C8H14O2/c1-3-5-6-7-10-8(9)4-2/h4H,2-3,5-7H2,1H3
<b>InchiKey:</b>	ULDDEWDFUNBUKM-UHFFFAOYSA-N
<b>Formula:</b>	C8H14O2
<b>SMILES:</b>	C=CC(=O)OCCCC
<b>Mol. weight [g/mol]:</b>	142.20
<b>CAS:</b>	2998-23-4

## Physical Properties

Property code	Value	Unit	Source
gf	-129.60	kJ/mol	Joback Method
hf	-327.82	kJ/mol	Joback Method
hfus	17.98	kJ/mol	Joback Method
hvap	41.89	kJ/mol	Joback Method
log10ws	-1.89		Crippen Method
logp	1.906		Crippen Method
mcvol	126.720	ml/mol	McGowan Method
pc	2775.92	kPa	Joback Method
rinpol	978.00		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	975.00		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	973.00		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	976.00		NIST Webbook
rinpol	975.00		NIST Webbook
rinpol	978.00		NIST Webbook
ripol	1273.00		NIST Webbook
ripol	1273.00		NIST Webbook
ripol	1265.00		NIST Webbook
ripol	1276.00		NIST Webbook

tb	455.41	K	Joback Method
tc	633.98	K	Joback Method
tf	250.32	K	Joback Method
vc	0.488	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	263.07	J/mol×K	455.41	Joback Method
cpg	274.71	J/mol×K	485.17	Joback Method
cpg	285.90	J/mol×K	514.93	Joback Method
cpg	296.66	J/mol×K	544.69	Joback Method
cpg	306.99	J/mol×K	574.46	Joback Method
cpg	316.89	J/mol×K	604.22	Joback Method
cpg	326.37	J/mol×K	633.98	Joback Method
dvisc	0.0029783	Paxs	250.32	Joback Method
dvisc	0.0015454	Paxs	284.50	Joback Method
dvisc	0.0009231	Paxs	318.68	Joback Method
dvisc	0.0006093	Paxs	352.87	Joback Method
dvisc	0.0004328	Paxs	387.05	Joback Method
dvisc	0.0003249	Paxs	421.23	Joback Method
dvisc	0.0002547	Paxs	455.41	Joback Method
hvapt	44.90	kJ/mol	382.50	NIST Webbook

## Sources

High-Pressure Phase Behavior for  
Pentyl Acrylate and Pentyl  
Acrylate in Supercritical Carbon  
Dioxide:  
McGowan Method:

<https://www.doi.org/10.1021/je060122s>

NIST Webbook:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

Crippen Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2998234&Units=SI>

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

cpg: Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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