

# Butanoic acid, 3-oxo-, 2-propenyl ester

<b>Other names:</b>	Acetoacetic acid, allyl ester Allyl acetoacetate Allyl acetylacetate
<b>Inchi:</b>	InChI=1S/C7H10O3/c1-3-4-10-7(9)5-6(2)8/h3H,1,4-5H2,2H3
<b>InchiKey:</b>	AXLMPTNTPOWPLT-UHFFFAOYSA-N
<b>Formula:</b>	C7H10O3
<b>SMILES:</b>	C=CCOC(=O)CC(C)=O
<b>Mol. weight [g/mol]:</b>	142.15
<b>CAS:</b>	1118-84-9

## Physical Properties

Property code	Value	Unit	Source
gf	-266.94	kJ/mol	Joback Method
hf	-419.76	kJ/mol	Joback Method
hfus	16.99	kJ/mol	Joback Method
hvap	46.41	kJ/mol	Joback Method
log10ws	-0.75		Crippen Method
logp	0.695		Crippen Method
mcvol	114.200	ml/mol	McGowan Method
pc	3306.75	kPa	Joback Method
tb	486.40	K	Joback Method
tc	677.00	K	Joback Method
tf	288.98	K	Joback Method
vc	0.439	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	238.50	J/molxK	486.40	Joback Method
cpg	248.34	J/molxK	518.17	Joback Method
cpg	257.76	J/molxK	549.93	Joback Method
cpg	266.77	J/molxK	581.70	Joback Method
cpg	275.37	J/molxK	613.47	Joback Method
cpg	283.57	J/molxK	645.24	Joback Method

cpg	291.36	J/molxK	677.00	Joback Method
dvisc	0.0025463	Paxs	288.98	Joback Method
dvisc	0.0014879	Paxs	321.88	Joback Method
dvisc	0.0009605	Paxs	354.79	Joback Method
dvisc	0.0006679	Paxs	387.69	Joback Method
dvisc	0.0004916	Paxs	420.59	Joback Method
dvisc	0.0003783	Paxs	453.50	Joback Method
dvisc	0.0003016	Paxs	486.40	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	467.70	K	98.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1118849&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1118849&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tbrp:** Boiling point at reduced pressure  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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