

# 24-Ethyl-31-nor-8,25-lanostadienol acetate

<b>Inchi:</b>	InChI=1S/C33H54O2/c1-10-24(11-2)13-12-22(3)25-16-20-33(9)27-14-15-28-30(5,6)29(3)
<b>InchiKey:</b>	PTEZIKVWEClONK-RGVKLLQESA-N
<b>Formula:</b>	C33H54O2
<b>SMILES:</b>	<chem>CC=C(CC)CCC(C)C1CCC2(C)C3=C(CCC12C)C1(C)CCC(OC(C)=O)C(C)(C)C1CC3</chem>
<b>Mol. weight [g/mol]:</b>	482.78

## Physical Properties

Property code	Value	Unit	Source
gf	210.40	kJ/mol	Joback Method
hf	-571.92	kJ/mol	Joback Method
hfus	39.89	kJ/mol	Joback Method
hvap	94.45	kJ/mol	Joback Method
log10ws	-10.20		Crippen Method
logp	9.440		Crippen Method
mvol	431.230	ml/mol	McGowan Method
pc	796.63	kPa	Joback Method
rinpol	3399.00		NIST Webbook
rinpol	3399.00		NIST Webbook
tb	1078.71	K	Joback Method
tc	1323.38	K	Joback Method
tf	662.63	K	Joback Method
vc	1.645	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1737.28	J/molxK	1078.71	Joback Method
cpg	1793.96	J/molxK	1119.49	Joback Method
cpg	1855.41	J/molxK	1160.27	Joback Method
cpg	1922.30	J/molxK	1201.05	Joback Method
cpg	1995.31	J/molxK	1241.82	Joback Method
cpg	2075.12	J/molxK	1282.60	Joback Method
cpg	2162.42	J/molxK	1323.38	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R110326&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R110326&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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