

# Propanoic acid, 3-bromo-2-chloro, 2-methylpropyl ester

Inchi:	InChI=1S/C7H12BrClO2/c1-5(2)4-11-7(10)6(9)3-8/h5-6H,3-4H2,1-2H3
InchiKey:	REXXRTDJURMHKN-UHFFFAOYSA-N
Formula:	C7H12BrClO2
SMILES:	CC(C)COC(=O)C(Cl)CBr
Mol. weight [g/mol]:	243.53

## Physical Properties

Property code	Value	Unit	Source
gf	-228.35	kJ/mol	Joback Method
hf	-432.58	kJ/mol	Joback Method
hfus	19.11	kJ/mol	Joback Method
hvap	50.38	kJ/mol	Joback Method
log10ws	-2.07		Crippen Method
logp	2.188		Crippen Method
mcvol	146.670	ml/mol	McGowan Method
pc	3100.18	kPa	Joback Method
rinpol	1214.00		NIST Webbook
rinpol	1214.00		NIST Webbook
tb	538.56	K	Joback Method
tc	744.15	K	Joback Method
tf	300.53	K	Joback Method
vc	0.550	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	302.24	J/molxK	538.56	Joback Method
cpg	351.47	J/molxK	709.89	Joback Method
cpg	342.70	J/molxK	675.62	Joback Method
cpg	333.40	J/molxK	641.36	Joback Method
cpg	323.57	J/molxK	607.09	Joback Method
cpg	313.19	J/molxK	572.83	Joback Method
cpg	359.73	J/molxK	744.15	Joback Method
dvisc	0.0002522	Paxs	538.56	Joback Method

dvisc	0.0003324	Paxs	498.89	Joback Method
dvisc	0.0004594	Paxs	459.22	Joback Method
dvisc	0.0006752	Paxs	419.54	Joback Method
dvisc	0.0010753	Paxs	379.87	Joback Method
dvisc	0.0019088	Paxs	340.20	Joback Method
dvisc	0.0039430	Paxs	300.53	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R30250&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R30250&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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