

# Phenyltoloxamine

**Other names:**

Ethanamine, N,N-dimethyl-2-[2-(phenylmethyl)phenoxy]-  
Ethylamine, N,N-dimethyl-2-[(«alpha»-phenyl-o-tolyl)oxy]-  
o-Benzylphenyl 2-(dimethylamino)ethyl ether  
Antin  
Bistrimin  
Bristamin  
C 5581 H  
Fenoxadrin  
Histionex  
Phenoxadrin  
Phenoxadrine  
Phentoloxamine  
PRN  
2-(2-Benzylphenoxy)-N,N-dimethylethylamine  
2-[2-(Dimethylamino)ethoxy]diphenylmethane  
2-Benzylphenyl-«beta»-dimethylamino ethyl ether  
N,N-Dimethyl-2-(«alpha»-phenyl-o-toloxyl)ethylamine  
N,N-Dimethyl-2-((«alpha»-phenyl-o-tolyl)oxy)ethylamine  
Phentoloxamin

**Inchi:**

InChI=1S/C17H21NO/c1-18(2)12-13-19-17-11-7-6-10-16(17)14-15-8-4-3-5-9-15/h3-11H,

**InchiKey:**

IZRPKIZLIFYYKR-UHFFFAOYSA-N

**Formula:**

C17H21NO

**SMILES:**

CN(C)CCOc1ccccc1Cc1ccccc1

**Mol. weight [g/mol]:**

255.35

**CAS:**

92-12-6

## Physical Properties

Property code	Value	Unit	Source
gf	313.23	kJ/mol	Joback Method
hf	2.69	kJ/mol	Joback Method
hfus	31.69	kJ/mol	Joback Method
hvap	63.10	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.218		Crippen Method
mcvol	218.720	ml/mol	McGowan Method
pc	2025.41	kPa	Joback Method
rinsol	1926.00		NIST Webbook

rinpol	1915.00		NIST Webbook
rinpol	1960.00		NIST Webbook
rinpol	1960.00		NIST Webbook
rinpol	1970.00		NIST Webbook
rinpol	1926.00		NIST Webbook
rinpol	1925.00		NIST Webbook
rinpol	1970.00		NIST Webbook
rinpol	1915.00		NIST Webbook
rinpol	1915.00		NIST Webbook
rinpol	1938.00		NIST Webbook
rinpol	1926.00		NIST Webbook
tb	681.56	K	Joback Method
tc	901.89	K	Joback Method
tf	401.41	K	Joback Method
vc	0.807	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	591.47	J/mol×K	681.56	Joback Method
cpg	609.79	J/mol×K	718.28	Joback Method
cpg	626.85	J/mol×K	755.00	Joback Method
cpg	642.71	J/mol×K	791.73	Joback Method
cpg	657.43	J/mol×K	828.45	Joback Method
cpg	671.07	J/mol×K	865.17	Joback Method
cpg	683.68	J/mol×K	901.89	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C92126&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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