

Tetracyano-p-quinodimethane

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|-----------------------------|--|
| Other names: | Propanedinitrile, 2,2'-(2,5-cyclohexadiene-1,4-diylidene)bis- 2,5-Cyclohexadiene-«delta»1,«alpha»:4,«alpha»'-dimalononitrile Quinodimethan, tetracyano- Tetracyanoquinodimethan(e) TCNQ 2,5-Cyclohexadiene-1,4-Diylidenedimalononitrile 7,7,8,8-Tetracyano-p-quinodimethane 7,7,8,8-Tetracyano-1,4-quinodimethan 7,7,8,8-Tetracyanoquinodimethane 7,7',8,8'-Tetracyanoquinodimethane 2,5-Cyclohexadiene-1,4-diylidene-«alpha»,«alpha»'-dimalononitrile Tetracyanoquinodimethan NSC 105237 2,2'-(2,5-cyclohexadiene-1,4-diylidene)bismalononitrile |
| Inchi: | InChI=1S/C12H4N4/c13-5-11(6-14)9-1-2-10(4-3-9)12(7-15)8-16/h1-4H |
| InchiKey: | PCCVSPMFGIFTHU-UHFFFAOYSA-N |
| Formula: | C12H4N4 |
| SMILES: | N#CC(C#N)=c1ccc(=C(C#N)C#N)cc1 |
| Mol. weight [g/mol]: | 204.19 |
| CAS: | 1518-16-7 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|-----------------|--------|----------------|
| chs | -5959.00 ± 3.00 | kJ/mol | NIST Webbook |
| chs | -5958.90 ± 5.70 | kJ/mol | NIST Webbook |
| ea | 2.80 ± 0.10 | eV | NIST Webbook |
| ea | 2.88 ± 0.19 | eV | NIST Webbook |
| ea | 2.80 ± 0.10 | eV | NIST Webbook |
| ea | 2.80 ± 0.30 | eV | NIST Webbook |
| hf | 771.10 | kJ/mol | NIST Webbook |
| hf | 770.00 ± 10.00 | kJ/mol | NIST Webbook |
| hfs | 665.00 ± 3.00 | kJ/mol | NIST Webbook |
| hfs | 666.10 ± 5.90 | kJ/mol | NIST Webbook |
| hsub | 105.00 | kJ/mol | NIST Webbook |
| hsub | 105.00 | kJ/mol | NIST Webbook |
| log10ws | -2.07 | | Crippen Method |
| logp | 0.082 | | Crippen Method |

| | | | |
|-------|---------------|--------|----------------|
| mcvol | 157.400 | ml/mol | McGowan Method |
| tf | 567.00 ± 2.00 | K | NIST Webbook |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|----------------|--------|-----------------|--------------|
| hsubt | 108.00 ± 2.00 | kJ/mol | 502.50 | NIST Webbook |
| hsubt | 122.00 ± 2.00 | kJ/mol | 423.00 | NIST Webbook |
| hsubt | 126.00 ± 1.00 | kJ/mol | 413.00 | NIST Webbook |
| hsubt | 105.00 ± 10.00 | kJ/mol | 466.00 | NIST Webbook |
| hsubt | 105.00 ± 9.20 | kJ/mol | 465.00 | NIST Webbook |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C1518167&Units=SI |

Legend

| | |
|-----------------|--|
| chs: | Standard solid enthalpy of combustion |
| ea: | Electron affinity |
| hf: | Enthalpy of formation at standard conditions |
| hfs: | Solid phase enthalpy of formation at standard conditions |
| hsub: | Enthalpy of sublimation at standard conditions |
| hsubt: | Enthalpy of sublimation at a given temperature |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| tf: | Normal melting (fusion) point |

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