

# 3-Heptene

<b>Other names:</b>	3-Heptene (c,t) hept-3-ene
<b>Inchi:</b>	InChI=1S/C7H14/c1-3-5-7-6-4-2/h5,7H,3-4,6H2,1-2H3
<b>InchiKey:</b>	WZHKDGSXCTSCCK-UHFFFAOYSA-N
<b>Formula:</b>	C7H14
<b>SMILES:</b>	CCC=CCCC
<b>Mol. weight [g/mol]:</b>	98.19
<b>CAS:</b>	592-78-9

## Physical Properties

Property code	Value	Unit	Source
chl	-4646.88 ± 0.71	kJ/mol	NIST Webbook
gf	88.28	kJ/mol	Joback Method
hf	-70.59	kJ/mol	Joback Method
hfus	14.09	kJ/mol	Joback Method
hvap	31.13	kJ/mol	Joback Method
ie	8.77 ± 0.02	eV	NIST Webbook
log10ws	-2.61		Crippen Method
logp	2.753		Crippen Method
mcvol	105.190	ml/mol	McGowan Method
pc	2960.12	kPa	Joback Method
rinpol	698.00		NIST Webbook
rinpol	698.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	708.30		NIST Webbook
tb	368.65 ± 2.00	K	NIST Webbook
tb	367.15 ± 2.00	K	NIST Webbook
tb	367.15 ± 2.00	K	NIST Webbook
tb	368.80 ± 2.00	K	NIST Webbook
tb	367.15 ± 2.00	K	NIST Webbook
tb	368.90 ± 1.00	K	NIST Webbook
tb	368.90 ± 1.00	K	NIST Webbook
tb	368.70 ± 1.00	K	NIST Webbook
tb	367.15 ± 2.00	K	NIST Webbook
tb	369.05 ± 1.00	K	NIST Webbook
tb	369.65 ± 2.00	K	NIST Webbook

tb	369.00 ± 1.00	K	NIST Webbook
tb	368.75 ± 0.70	K	NIST Webbook
tb	368.90 ± 0.50	K	NIST Webbook
tb	368.95 ± 1.00	K	NIST Webbook
tb	368.65 ± 2.00	K	NIST Webbook
tb	369.20 ± 1.00	K	NIST Webbook
tb	369.20 ± 1.00	K	NIST Webbook
tb	369.25 ± 1.00	K	NIST Webbook
tb	370.65 ± 2.00	K	NIST Webbook
tc	536.01	K	Joback Method
tf	163.57	K	Joback Method
vc	0.407	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	180.32	J/molxK	363.72	Joback Method
cpg	233.95	J/molxK	507.29	Joback Method
cpg	224.15	J/molxK	478.58	Joback Method
cpg	213.90	J/molxK	449.86	Joback Method
cpg	203.19	J/molxK	421.15	Joback Method
cpg	192.00	J/molxK	392.43	Joback Method
cpg	243.33	J/molxK	536.01	Joback Method
dvisc	0.0001967	Paxs	363.72	Joback Method
dvisc	0.0002558	Paxs	330.36	Joback Method
dvisc	0.0003528	Paxs	297.00	Joback Method
dvisc	0.0005278	Paxs	263.64	Joback Method
dvisc	0.0008875	Paxs	230.29	Joback Method
dvisc	0.0017796	Paxs	196.93	Joback Method
dvisc	0.0047393	Paxs	163.57	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49153e+01
Coeff. B	-3.34928e+03

Coeff. C	-4.37320e+01
Temperature range (K), min.	272.70
Temperature range (K), max.	392.48

## Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C592789&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C592789&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/30-726-2/3-Heptene.pdf>

Generated by Cheméo on 2024-04-18 04:39:03.126509447 +0000 UTC m=+15704392.047086769.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.