

# Butanoic acid, 3-methyl, 2-iodoethyl ester

Inchi:	InChI=1S/C7H13IO2/c1-6(2)5-7(9)10-4-3-8/h6H,3-5H2,1-2H3
InchiKey:	ANMUQHGRSUBSQJ-UHFFFAOYSA-N
Formula:	C7H13IO2
SMILES:	CC(C)CC(=O)OCCI
Mol. weight [g/mol]:	256.08

## Physical Properties

Property code	Value	Unit	Source
gf	-170.18	kJ/mol	Joback Method
hf	-361.02	kJ/mol	Joback Method
hfus	17.56	kJ/mol	Joback Method
hvap	49.32	kJ/mol	Joback Method
log10ws	-2.32		Crippen Method
logp	2.011		Crippen Method
mcvol	142.750	ml/mol	McGowan Method
pc	2896.73	kPa	Joback Method
rinpol	1203.00		NIST Webbook
tb	528.55	K	Joback Method
tc	739.58	K	Joback Method
tf	283.87	K	Joback Method
vc	0.533	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.34	J/molxK	528.55	Joback Method
cpg	299.75	J/molxK	563.72	Joback Method
cpg	310.60	J/molxK	598.89	Joback Method
cpg	320.88	J/molxK	634.07	Joback Method
cpg	330.61	J/molxK	669.24	Joback Method
cpg	339.81	J/molxK	704.41	Joback Method
cpg	348.49	J/molxK	739.58	Joback Method
dvisc	0.0046711	Paxs	283.87	Joback Method
dvisc	0.0021715	Paxs	324.65	Joback Method

dvisc	0.0011976	Paxs	365.43	Joback Method
dvisc	0.0007444	Paxs	406.21	Joback Method
dvisc	0.0005046	Paxs	446.99	Joback Method
dvisc	0.0003650	Paxs	487.77	Joback Method
dvisc	0.0002776	Paxs	528.55	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R19854&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R19854&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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