

# 2-Octene, (E)-

<b>Other names:</b>	(2E)-2-Octene (E)-2-C <sub>8</sub> H <sub>16</sub> (E)-2-OCTENE (E)-Oct-2-ene .beta.-trans-octene Oct-2(E)-ene TRANS-2-OCTENE trans-oct-2-ene
<b>Inchi:</b>	InChI=1S/C <sub>8</sub> H <sub>16</sub> /c1-3-5-7-8-6-4-2/h3,5H,4,6-8H2,1-2H3/b5-3+
<b>InchiKey:</b>	ILPBINAXDRFYPL-HWKANZROSA-N
<b>Formula:</b>	C <sub>8</sub> H <sub>16</sub>
<b>SMILES:</b>	CC=CCCCC
<b>Mol. weight [g/mol]:</b>	112.21
<b>CAS:</b>	13389-42-9

## Physical Properties

Property code	Value	Unit	Source
af	0.3500		KDB
gf	92.74	kJ/mol	KDB
hcg	5315.77	kJ/mol	KDB
hcn	4963.898	kJ/mol	KDB
hf	-94.58	kJ/mol	KDB
hfus	16.68	kJ/mol	Joback Method
hvap	40.20	kJ/mol	NIST Webbook
ie	8.91 ± 0.01	eV	NIST Webbook
ie	9.09 ± 0.01	eV	NIST Webbook
log10ws	-3.02		Crippen Method
logp	3.143		Crippen Method
mcvol	119.280	ml/mol	McGowan Method
pc	2770.00	kPa	KDB
rinpola	803.00		NIST Webbook
rinpola	815.00		NIST Webbook
rinpola	810.00		NIST Webbook
rinpola	804.00		NIST Webbook
rinpola	805.00		NIST Webbook
rinpola	804.00		NIST Webbook
rinpola	798.00		NIST Webbook

rinpol	817.00	NIST Webbook
rinpol	807.00	NIST Webbook
rinpol	818.00	NIST Webbook
rinpol	818.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	810.00	NIST Webbook
rinpol	811.00	NIST Webbook
rinpol	809.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	816.00	NIST Webbook
rinpol	815.00	NIST Webbook
rinpol	801.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	803.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	797.40	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	799.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	803.00	NIST Webbook
rinpol	803.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	804.00	NIST Webbook
rinpol	806.00	NIST Webbook
rinpol	797.00	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	792.00	NIST Webbook
rinpol	797.50	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	798.20	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	797.00	NIST Webbook
rinpol	798.00	NIST Webbook
rinpol	803.60	NIST Webbook
rinpol	797.20	NIST Webbook
rinpol	797.00	NIST Webbook
rinpol	797.30	NIST Webbook

ripol	797.70		NIST Webbook
ripol	799.00		NIST Webbook
ripol	805.00		NIST Webbook
ripol	805.00		NIST Webbook
ripol	815.00		NIST Webbook
ripol	797.90		NIST Webbook
ripol	810.00		NIST Webbook
ripol	803.00		NIST Webbook
ripol	805.00		NIST Webbook
ripol	864.00		NIST Webbook
ripol	866.00		NIST Webbook
ripol	853.00		NIST Webbook
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ripol	852.00		NIST Webbook
ripol	853.00		NIST Webbook
ripol	866.00		NIST Webbook
ripol	866.00		NIST Webbook
ripol	866.50		NIST Webbook
ripol	864.30		NIST Webbook
ripol	865.70		NIST Webbook
ripol	864.00		NIST Webbook
ripol	866.50		NIST Webbook
ripol	864.00		NIST Webbook
ripol	864.30		NIST Webbook
ripol	865.70		NIST Webbook
ripol	866.00		NIST Webbook
ripol	864.00		NIST Webbook
ripol	865.70		NIST Webbook
ripol	860.00		NIST Webbook
tb	398.26 ± 0.20	K	NIST Webbook
tb	398.09 ± 0.30	K	NIST Webbook
tb	398.35 ± 0.50	K	NIST Webbook
tb	397.40 ± 0.60	K	NIST Webbook
tb	397.40 ± 0.60	K	NIST Webbook
tb	398.10	K	KDB
tb	396.20	K	NIST Webbook
tb	398.09 ± 0.30	K	NIST Webbook
tb	398.20	K	NIST Webbook
tb	396.00	K	NIST Webbook
tb	396.50 ± 1.00	K	NIST Webbook

tc	580.00	K	KDB
tc	569.80	K	Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons
tf	185.15 ± 1.50	K	NIST Webbook
tf	185.40	K	KDB
vc	0.464	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	287.62	J/mol×K	558.45	Joback Method
cpg	218.05	J/mol×K	386.60	Joback Method
cpg	230.94	J/mol×K	415.24	Joback Method
cpg	243.29	J/mol×K	443.88	Joback Method
cpg	255.11	J/mol×K	472.53	Joback Method
cpg	266.43	J/mol×K	501.17	Joback Method
cpg	277.26	J/mol×K	529.81	Joback Method
dvisc	0.0002015	Paxs	386.60	Joback Method
dvisc	0.0052396	Paxs	174.84	Joback Method
dvisc	0.0019294	Paxs	210.13	Joback Method
dvisc	0.0009470	Paxs	245.43	Joback Method
dvisc	0.0005559	Paxs	280.72	Joback Method
dvisc	0.0003675	Paxs	316.01	Joback Method
dvisc	0.0002641	Paxs	351.31	Joback Method
hvapt	37.90	kJ/mol	377.50	NIST Webbook
rfi	1.41070		298.15	KDB
rhol	720.00	kg/m <sup>3</sup>	293.00	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47525e+01
Coeff. B	-3.56590e+03
Coeff. C	-4.14850e+01
Temperature range (K), min.	288.01

Temperature range (K), max.	419.19
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Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	6.58492e+01
Coeff. B	-6.79412e+03
Coeff. C	-7.48209e+00
Coeff. D	3.98341e-06
Temperature range (K), min.	185.45
Temperature range (K), max.	577.00

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=251">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=251</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=251">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=251</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13389429&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13389429&amp;Units=SI</a>
<b>Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons:</b>	<a href="https://www.doi.org/10.1021/je0341357">https://www.doi.org/10.1021/je0341357</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>af:</b>	Acentric Factor
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hcg:</b>	Heat of Combustion, Gross form
<b>hcn:</b>	Heat of Combustion, Net Form
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rfi:</b>	Refractive Index
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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